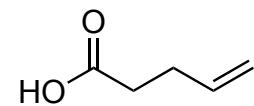


Synthesis Challenge #10 AG Wegner

JLU Giessen

16.01.2014



A

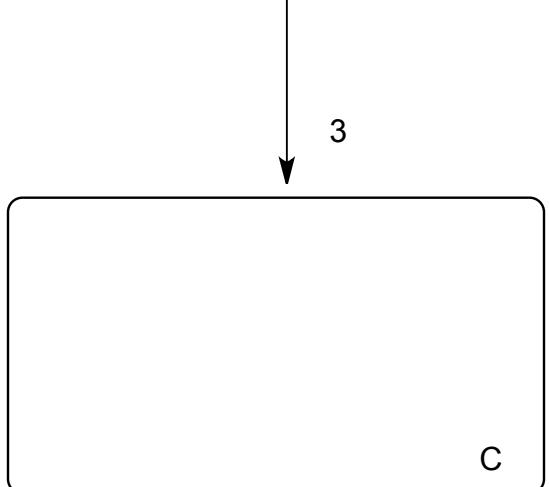
↓  
1-2



B

1)  $\text{H}_2\text{NOMe}^*\text{HCl}$ , EDCI, HOEt,  
 $\text{Et}_3\text{N}$   
2) Grubbs II,  $\text{CH}_2=\text{CHCH}_2\text{TMS}$

Please, provide a detailed Mechanism for step 1).



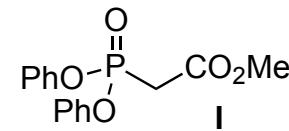
3

3)  $\text{Br}(\text{CH}_2)_3\text{CHO}$ ,  $\text{BF}_3^*\text{Et}_2\text{O}$

Please, provide a detailed mechanism for step 3).

↓  
4-5

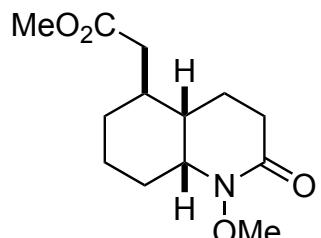
4) O<sub>3</sub>, MeOH, -78°C, Me<sub>2</sub>S, RT  
5) I, DBU, NaI



D

↓  
6

6) AIBN, Bu<sub>3</sub>SnH, Toluene, 80°C



E

↓  
7

7) [Cp<sub>2</sub>ZrHCl], CH<sub>2</sub>Cl<sub>2</sub>, 0°C;  
CH<sub>2</sub>=CHCH<sub>2</sub>SnBu<sub>3</sub>, cat.  
Sc(OTf)<sub>3</sub>

Please, provide mechanism for step 7)

F

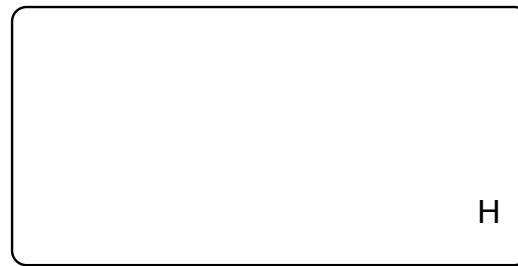
↓ 8-9



- 8) thexylborane, THF, 0°C;  
NaBO<sub>3</sub>, RT  
9) SO<sub>3</sub>\*Py, *i*Pr<sub>2</sub>NEt,  
DMSO/CH<sub>2</sub>Cl<sub>2</sub>, RT;  
Ph<sub>3</sub>P=CHCO<sub>2</sub>*t*Bu

What is the name of the reaction in step 9)?

↓ 10-12

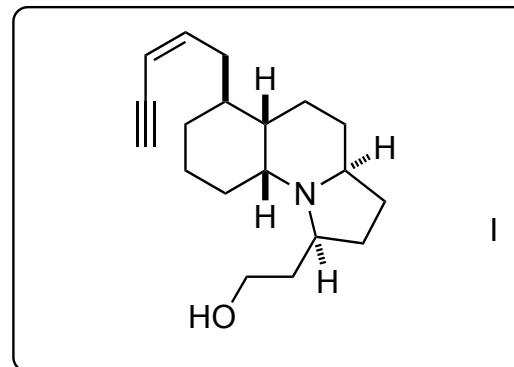


- 10) Zn, AcOH/H<sub>2</sub>O  
11) NaAlH(O*t*Bu)*i*Bu<sub>2</sub>  
12) Ph<sub>3</sub>P<sup>+</sup>CH<sub>2</sub>I<sup>-</sup>, NaN(TMS)<sub>2</sub>  
THF/HMPA, -78°C to RT



II

↓ 13-14



- 13) II, [Pd(PPh<sub>3</sub>)<sub>4</sub>], CuI, Et<sub>3</sub>N, RT;  
K<sub>2</sub>CO<sub>3</sub>, MeOH, RT  
14) DIBALH, Et<sub>2</sub>O, -78°C;  
NaBH<sub>4</sub>, MeOH

What is the name of the reaction in step 13?  
What is the purpose of CuI?