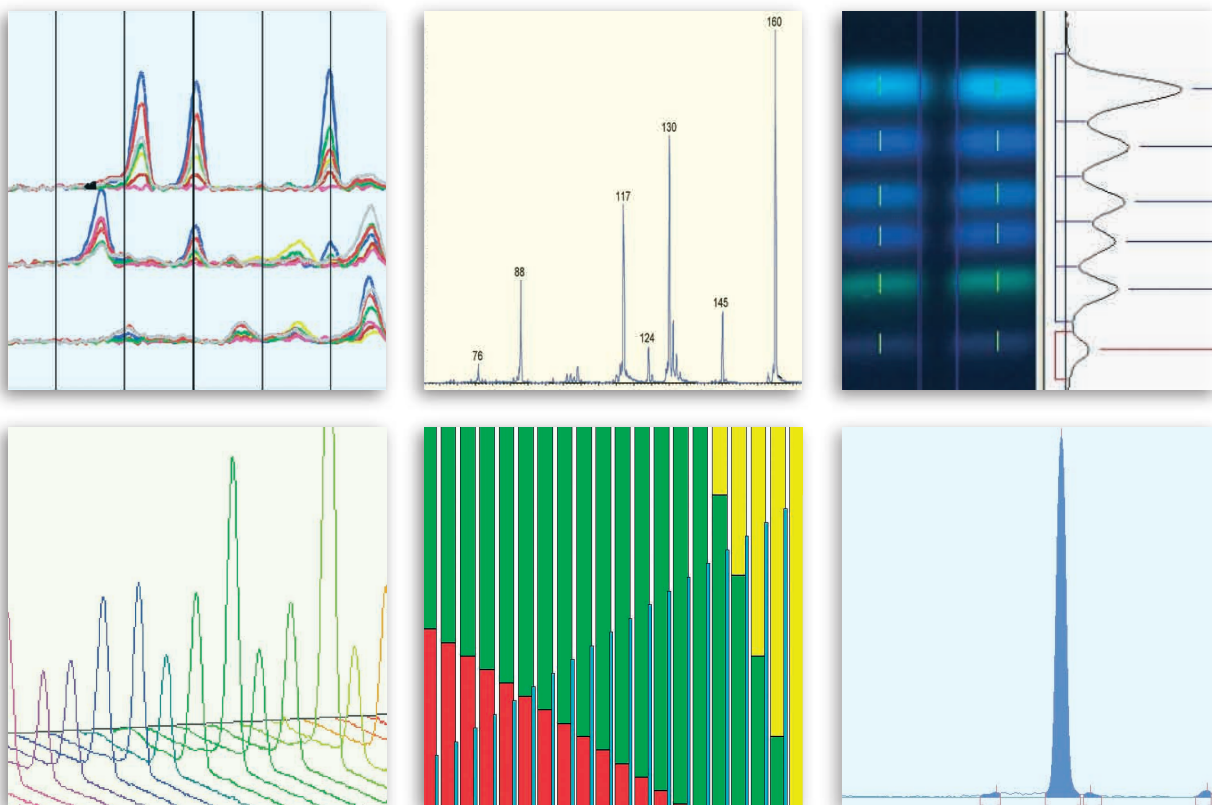


# CBS

CAMAG BIBLIOGRAPHY SERVICE



**Planar chromatography – an essential component of modern analysis**

**CAMAG**

**98**

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# Planar Chromatography in Practice

## Determination of amino-propanol in dermatological products



▲ Bayer Santé Familiale, Gaillard, France



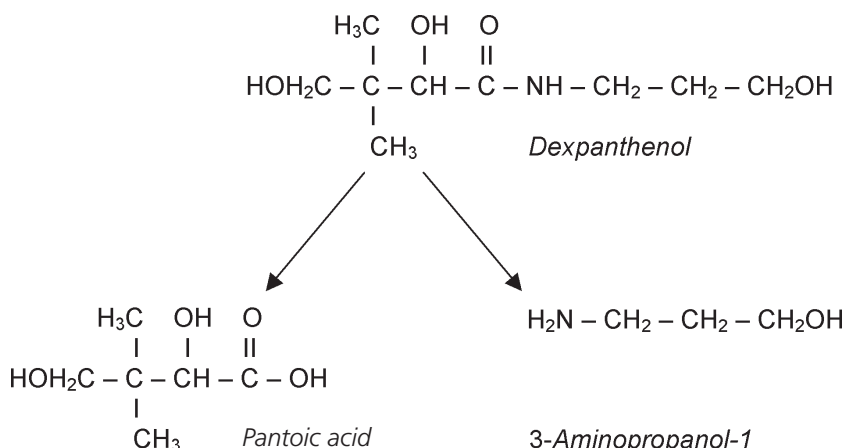
▲ Caroline Petitti, analytical development chemist

The Analytical Development Laboratory of Bayer Consumer Care International Technical Center in Gaillard, France, develops methods for analysis of OTC (over the counter) products. Most of the analyses are performed by HPLC, AAS (atomic absorption spectrophotometry), ICP-OES (inductively coupled plasma optical emission spectrometry) and potentiometry, and some by UV or IR spectroscopy, GC and HPTLC.

**Amino-propanol is usually analyzed by HPLC, however, this method is very time consuming. Therefore, it has been replaced by a simple, rapid and accurate HPTLC method. The very good performance is proven by the data of validation (correlation of the calibration function 0.9979, mean recovery 102 % ±4.9 %, intermediate precision ±5.7 %). The main advantage is increased sample-throughput, i.e. by a factor of 3 with a significant reduction in working time.**

### Introduction

Amino-3-propan-1-ol is the degradation product of the dexpanthenol (pro vitamin B5), which is one of the active ingredients in the Bayer dermatological products Bepanthen® and Bepanthol®. Dexpanthenol degrades into pantoic acid and amino-propanol according to the scheme below. The stability control of dexpanthenol is performed by determination of the amino propanol content, whereby a high sample throughput is given.



## Sample preparation

Cream or ointment was dissolved in ethanol, followed by heating. Insoluble matter was separated by centrifugation at 10 000 rpm for 5 min.

## Standard preparation

Amino-propanol was dissolved and diluted in ethanol in order to obtain a five-point calibration ranged between 12.5 µg/mL and 200 µg/mL.

*Note by the editor: When the spray-on technique is used, different amounts (volumes) of one calibration solution can be applied.*

## Chromatogram layer

HPTLC plate silica gel 60 F<sub>254</sub> (Merck), pre-washed with methanol (pre-chromatography) and dried at 105 °C for 30 min on the TLC plate heater III.

## Sample application

As 6 mm bands with the Automatic TLC Sampler, max. 23 tracks, track distance min. 7.8 mm, distance from lower edge of plate 8 mm, distance from the sides at least 14 mm, application volume 2 µL for each calibration solution and sample.

## Chromatography

In the mobile phase mixture of ethanol – water – acetic acid 16:3:1 (v/v/v) the derivatization reagent was added, i.e. 0.5 g ninhydrin were dissolved in 100 mL solvent mixture. Chromatography was performed in the horizontal developing chamber from both plate sides up to migration distance of 40 mm (from plate edge).

## Derivatization

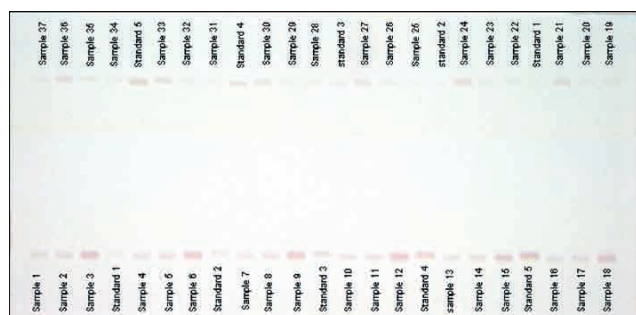
The plate was heated at 105 °C for 5 min on the TLC plate heater III and of amino-propanol was visible as a light-violet zone at  $hR_F$  50.

## Densitometric evaluation

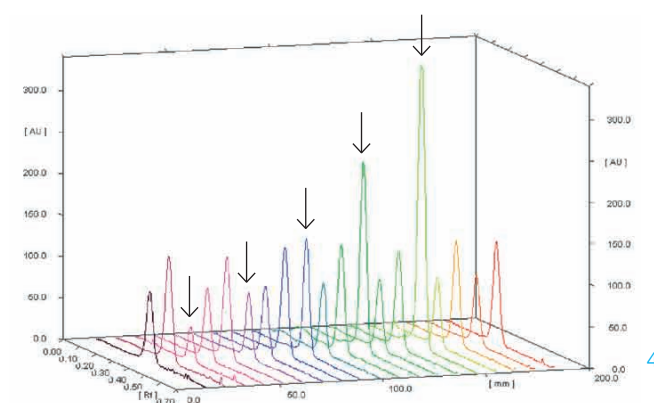
Absorbance measurement at 486 nm by TLC Scanner 3 and winCATS software; evaluation via peak area by Michaelis Menten 2 regression.

## Documentation

With Reprstar 3 (reflectance mode) under white light illumination.



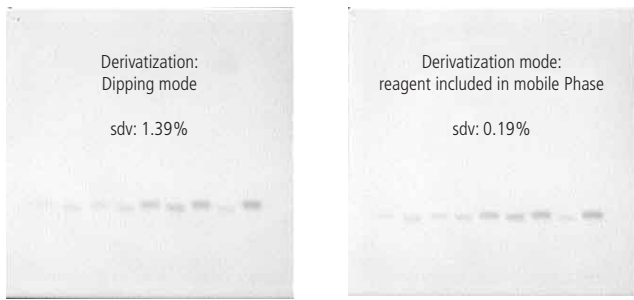
▲ Example of a HPTLC plate with the violet amino-propanol zone; respective sample and standard tracks were developed from both plate sides in the HDC and just heated for derivatization



▲ 3D-graphic of the close-up range of the amino-propanol peak on different tracks measured at 486 nm (five-point calibration is marked)

## Results and discussion

This HPTLC method is a good alternative to the HPLC method, since the protocol is simpler and faster, allowing up to 36 samples to be analyzed simultaneously. Another significant advantage is the ease of derivatization: chromatography and derivatization were performed at the same time by just adding the derivatization reagent ninhydrin into the mobile phase mixture. By this very homogenous application of the derivatization reagent, the relative standard deviation of the calibration function (sdv) was even improved (sdv ±0.19 %) compared to the already very good precision obtained by dipping (sdv ±1.39 %).



▲ Comparison of derivatization modes: the relative standard deviation of the calibration function (sdv) is improved by adding the reagent into the mobile phase from  $sdv \pm 1.39\%$  to  $sdv \pm 0.19\%$

For the analysis of 30 samples the HPLC method required 3 days whereas the HPTLC method took only 1 day. By using HPTLC the analysis throughput was increased by factor 3.

Method used	HPLC	HPTLC
Time required for 30 Samples	3 Days	1 Day

The perfect suitability of HPTLC for this analysis was evident in the validation data. The limit of detection and quantification (LOD and LOQ) was  $4.5 \mu\text{g/mL}$  and  $15 \mu\text{g/mL}$ , respectively, related to the applied solution volumes of  $2 \mu\text{L}$  (if necessary LOD and LOQ can be improved by application of higher volumes.). The correlation coefficient of the calibration function was 0.9979. Accuracy was expressed by the mean recovery ( $n=15$ ) which was 102 %. The mean repeatability was established to be  $\pm 4.9\%$  at 5 different concentration levels ( $n=3$  for each calibration level). The relative standard deviation of the intermediate precision (RSD,  $n=9$ ) was  $\pm 5.7\%$ .

Validation parameter	Results
LOD	$4.5 \mu\text{g/mL}$
LOQ	$15 \mu\text{g/mL}$
<i>(both related to the applied volume <math>2 \mu\text{L}</math>)</i>	
Linearity <i>(coefficient of correlation)</i>	0.9979
Mean recovery	102 %
Mean repeatability (RSD)	$\pm 4.9\%$
<i>(performed at 5 concentration levels with <math>n=3</math> for each calibration level)</i>	
Intermediate precision (RSD, $n=9$ )	$\pm 5.7\%$

Further information is available from the author on request.

\*Caroline Petitti, Bayer Santé Familiale, 33 rue de l'industrie, F-74 240 Gaillard, France, caroline.petitti@bayerhealthcare.com

## Stability testing of gatifloxacin and analysis in polymeric nanoparticles



▲ Ms. Shruti Chopra and Mr. Sanjay Motwani

In CBS 97 we introduced the Indian HPTLC group from the Faculty of Pharmacy, Jamia Hamdard in New Delhi, India which is actively involved in the development and validation of HPTLC methods for drugs and biomarkers from herbal sources. It is well known that the pharmaceutical industry needs fast, reliable and at the same time economic methods to ensure drugs content in their different pharmaceutical formulations. Therefore the development of high-throughput techniques is highly appreciated, where HPTLC, among other methods, is an excellent tool for quality control under GMP regulations.

**In India HPTLC has become a part of routine analytical techniques in many product development and analytical laboratories. HPTLC proves to be more economical for analysis of pharmaceutical dosage forms than other chromatographic methods as it utilizes smaller volumes of solvents and a minimum sample clean up. Many samples can be simultaneously analyzed in a short time using automated devices.**

In this work the group of Sanjay Motwani\* developed and validated according to ICH guidelines a precise and robust HPTLC method [1] for gatifloxacin stability tests and for analysis of gatifloxacin in polymeric nanoparticles.

### Introduction

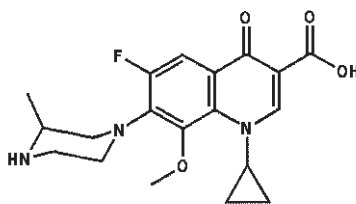
Gatifloxacin is a novel broad-spectrum fluoroquinolone widely used in USA, however restricted in use for Europe. It inhibits bacterial DNA gyrase

and topoisomerase IV and is effective in a range of clinical infections, with improved activity against gram-positive and gram-negative aerobic bacteria and atypical bacteria.

The International Conference on Harmonization (ICH) guidelines Q1A (R2) entitled 'Stability testing of new drug substances and products' emphasize that the stress testing of the drug substance should be carried out to elucidate the inherent stability characteristics of the active drug substance. Susceptibility to oxidation is one of the required tests, besides acid or alkaline hydrolysis and photolytic stability studies. An ideal stability-indicating method shall quantify the drug and also resolve its degradation products. Analytical methods for gatifloxacin include fluorimetry, HPLC, HPLC/ESI-MS(-MS), and HPTLC. However, no HPTLC method related to the stability test of gatifloxacin and its determination from polymeric nanoparticles was reported so far. Thus the following HPTLC method was developed which fulfills the above mentioned criteria completely.

### Sample preparation

Gatifloxacin bulk drug was dissolved in methanol (1 mg/mL). Polymeric polybutylcyanoacrylate nanoparticles (label claim: 2.48 mg of gatifloxacin per 100 mg of nanoparticles) equivalent to 5.5 mg of gatifloxacin were accurately weighed and transferred into a volumetric flask containing 5 mL acetone. To ensure complete extraction, it was sonicated for 30 min, diluted to 10 mL with acetone and centrifuged at 2000 rpm for 5 min. After filtration, the supernatant was analyzed.



▲ Structure formula of gatifloxacin [1-cyclopropyl-6-fluoro-1,4-dihydro-8-methoxy-7-(3-methyl-1-piperazinyl)-4-oxo-3-quinoline carboxylic acid]

## Standard solutions

For determination of the functional correlation gatifloxacin was dissolved in methanol (100 µg/mL). For analysis of related impurities gatifloxacin was dissolved in methanol (1 mg/mL) and diluted 1:5 with methanol (200 µg/mL).

## Layer

HPTLC aluminum foil silica gel 60 F<sub>254</sub> (Merck) 20 × 10 cm, pre-washed with methanol (by development) and dried at 90 °C for 30 min

## Sample application

Bandwise with Linomat V, application volume 2 µL for samples and 4–12 µL for the standard solution (400–1200 ng/band), for related impurity analysis 3 µL of sample and 2 µL of standard solution, band length 6 mm, track distance 10 mm, distance from lower edge 10 mm and distance from both sides at least 15 mm

## Chromatography

In a twin trough chamber pre-saturated for 30 min with the mobile phase n-propanol – methanol – ammonia (25 %) 5:1:0.9 (v/v/v). The migration distance was 80 mm from the lower plate edge. After chromatography, the plates were dried in a stream of warm air for 1 min.

## Densitometry

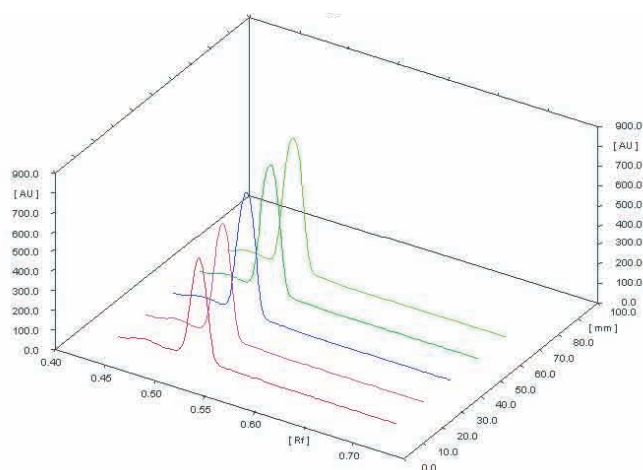
In absorbance mode at 292 nm using TLC Scanner 3 with winCATS software, linear calibration by peak area.

## Results and discussion

The ammonia content was important in the mobile phase to obtain a sharp and well-defined symmetrical gatifloxacin peak at  $hR_f$   $60 \pm 2$ . The regression analysis (mean of 3 calibration curves) showed good linear relationship with a correlation coefficient of 0.9954 in the concentration range of 400–1200 ng/band.

Calibration data of in the range of 400–1200 ng/band ( $n = 3$ )

Correlation coefficient ( $r \pm SD$ )	0.9954 ± 0.0012
Slope ± SD	9.6638 ± 0.0491
Confidence limit (95%) of slope	9.516 - 9.760
Intercept ± SD	956.3310 ± 27.6714
Confidence limit (95%) of intercept	936.26 - 987.90



▲ 3-D graphic of the calibration standard tracks in the range of 400–1200 ng/band, absorbance measurement at 292 nm

The repeatability of the measurement of the peak areas ( $n = 6$ ) at three different concentration levels (600, 800 and 1000 ng/band) showed low values of standard error  $SE < \pm 0.8$  and relative standard deviations  $RSD < \pm 0.03$  % for intra-day variation and  $SE < \pm 1.0$  and  $RSD < \pm 0.03$  % for inter-day variation, which is an excellent precision of the method. After introducing small deliberate changes in the critical parameters of the developed HPTLC method, the still low  $RSD$  ( $n = 3$ , 800 ng/band) of  $< \pm 0.03$  % and  $SE$  of  $\leq \pm 0.8$  indicated the robustness of the method. Limits of detection and quantification were 2.7 ng ( $S/N$  3) and 8.3 ng ( $S/N$  10), respectively, indicating an very adequate sensitivity of the method.

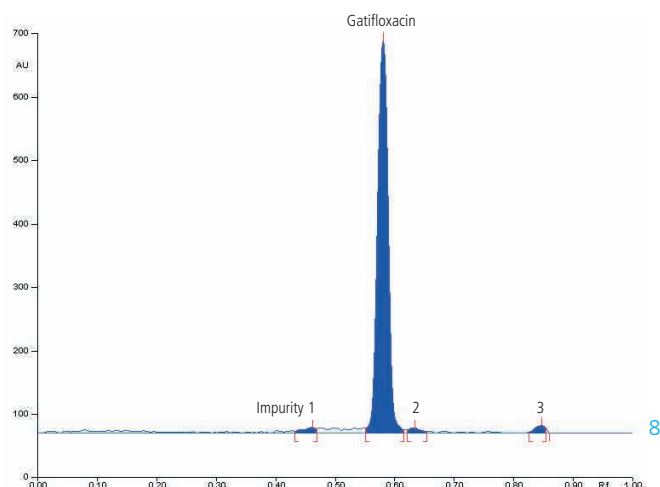
For recovery studies, 3 different amounts of gatifloxacin corresponding to 50, 100 and 150 % of the gatifloxacin label claim of polymeric nanoparticles were added. At each level, six determinations were performed. The recoveries obtained after extraction of the polymeric nanoparticles ranged between 99.2 and 101.9 %.

Moreover, the HPTLC method was perfectly suited to study parameters under different stress conditions with possible impact on degradation as per the recommendations of ICH guidelines. Standards for potential degradation products were not commercially available. As indicator for the gatifloxacin stability the recovery rate was used. Because of the reduced recovery rate under the influence of moist heat, two parameters of influence became evident. For the other five stress conditions the recovery rate was 99 %.

No.	Exposure conditions	Time (h)	$hR_F$ value of degradation products	Recovery rate (%)
1	Acid, 5 M HCl, refluxed	3	1, 3, 51, 75	46.1
2	Base, 5 M NaOH, refluxed	3	1	98.9
3	H <sub>2</sub> O <sub>2</sub> (30%, v/v), refluxed	3	None detected	99.2
4	Dry heat at 100 °C	8	54	98.5
5	Wet heat at 100 °C	3	52	97.7
6	Photostability – Daylight	24	None detected	99.3
7	UV 254 nm	8	None detected	99.0

The identity of the gatifloxacin peak was assessed by comparison of sample with standard spectra and a good correlation of 0.9996 was obtained between both. Peak purity was ascertained by spectra comparison within a peak at different positions (peak start, peak apex, peak end). There was no interference from excipients and other active components present in the polymeric nanoparticulate formulation.

The bulk drug sample (3000 ng gatifloxacin applied) for impurity analysis showed three additional peaks at  $hR_F$  values of  $46 \pm 2$ ,  $63 \pm 1$  and  $84 \pm 2$ . However, the area of the additional bands (altogether peak areas were 359 AU) was found to be significantly smaller than the peak area of the gatifloxacin standard solution (400 ng/band, peak area 9006 AU) and thus < 0.1 %.



▲ Chromatogram of gatifloxacin (3 µg/band) and its related impurities; all potential impurities at  $hR_F$  values of 46, 63 and 84 were all in all < 0.1 %

The developed and validated HPTLC method proved to be an accurate, precise, specific and reproducible method for the determination of gatifloxacin as a bulk drug and from pharmaceutical formulations.

Further information is available from the authors on request.

[1] S. K. Motwani, R. K. Khar, F. J. Ahmad, S. Chopra, K. Kohli, S. Talegaonkar, Z. Iqbal, *Analytica Chimica Acta* 576 (2006) 253–260

\*Sanjay K. Motwani, Department of Pharmaceutics, Faculty of Pharmacy, Jamia Hamdard, Hamdard Nagar, New Delhi – 110 062, Indien, Tel.: +91-11-26059688, Fax: +91-11-26059663, sanjay\_bcp@rediffmail.com

## Change in the management of Sales & Marketing



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Mr. Erwin Malzacher has been with CAMAG since November 1971 and has been responsible for sales and marketing since the end 1994. As part of his retirement plan, he reduced his work load to 60 % by mid 2005. We considered this task too important to CAMAG for it to be handled completely by a part time person, so in 2005 we began the search for a successor, allowing sufficient time for the person to grow into this position.

We are very happy to have found Ms Claudia Nachbur, a very competent and highly motivated person for this function. She joined CAMAG in November 2005 and, effective 1.1.2007, we appointed her Manager of Sales & Marketing. Erwin Malzacher who took great effort to introduce Claudia Nachbur to the CAMAG corporate culture and our market will continue to be available for our Sales & Marketing department with his invaluable experience of long standing.



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Claudia Nachbur studied at the ETH Zürich from 1982 to 87 and finished with a diploma of Natural Sciences. Before joining CAMAG she acquired many years of experience as sales manager of two Swiss companies in the international business with investment goods, including set up and support of the respective sales organizations.

Already during her first year with CAMAG Claudia Nachbur successfully managed several international missions. These included a customer seminar in Tashkent, Uzbekistan in March 2006 and participation at the 9th Colacro Conference in Merida, Mexico in June 2006 after which she visited several of our distributors in Central America.



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◀ Claudia Nachbur at the seminar at the Institute of Chemistry of Natural Plants in Tashkent



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◀ Participants and lecturers of the HPTLC workshop at the Colacro Conference – on the right Claudia Nachbur.

Meanwhile Claudia Nachbur is now in full control of our Sales & Marketing department. She has made several trips to Eastern European countries to meet and work with our distributors and she is currently busy with the reorganization and streamlining of our distributors network.



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CAMAG BIBLIOGRAPHY SERVICE  
PLANAR CHROMATOGRAPHY**

# CBS

Liebe Freunde

HPTLC-Symposia im internationalen Massstab haben eine lange Tradition. Das erste »International Symposium on Instrumental High Performance Thin-Layer Chromatography« fand 1979 in Bad Dürkheim statt mit Vortragenden aus 7 Nationen – damals ein beachtliche Internationalität! Das jüngste Symposium im Oktober letzten Jahres lockte über 150 Teilnehmer aus 22 Ländern mit Vortragenden aus 10 Nationen nach Berlin.

Eine erfreuliche Vielzahl jüngerer Teilnehmer sowohl aus dem universitären wie auch aus dem industriellen Bereich präsentierte beeindruckend über 50 Poster, die den Facettenreichtum der HPTLC widerspiegeln.

Nach einem Workshop gab es Präsentationen zu Grundlagen und neuen Aspekten, aber auch hervorragende Anwendungen in der Lebensmittelanalytik, Analytik von pflanzlichen Extrakten und klinischen Analytik. Den Abschluss bildeten innovative Entwicklungen in der HPTLC, Bioaktivitätstests, spezifische Immuno-Detektionen und Kopplungsmethoden. Die Internationalität der Beiträge in den 6 Themenblöcken verdeutlichte die weltweite Aktualität und zeigte das quo vadis der Methode auf.

Wünschen wir der Planar-Chromatographie auch in Zukunft solche Impulse. Für diejenigen, die nicht dabei sein konnten, finden sich Eindrücke unter [www.hptlc.com](http://www.hptlc.com) – es lohnt sich, da reinzuschauen... übrigens der erste Beitrag in dieser CBS-Ausgabe wurde auf dem Symposium eindrücklich präsentiert. Erfreulich, dass im boomenden Gesundheitsbereich auf die HPTLC rückbesonnen wird.

Herzlichst Ihre

*Gerda Morlock*

Gerda Morlock  
cbs@camag.com

Dear friends

HPTLC symposia on an international platform have a long tradition. The first "International Symposium on Instrumental High Performance Thin-Layer Chromatography" took place in Bad Dürkheim 1979 with scientific presentations from scientists from 7 nations all over the world – at that time a respectable international representation.



The latest symposium in Berlin in October last year brought together over 150 scientists from 22 countries and lecturers from 10 nations.

Many young attendees from academia as well as from industry participated, demonstrating that planar chromatography is very much alive and over 50 posters were presented, reflecting a myriad of features of HPTLC.

Presentations were given to fundamentals and new aspects, but also outstanding applications in food analysis, herbal analysis and clinical analysis. The closure part was dedicated to innovative developments in HPTLC, bioactivity tests, specific immuno detection and coupling methods. The internationality of the presentations in the 6 sessions underlined the worldwide up-to-dateness and showed the "quo vadis" of the technique.

We wish planar chromatography more such impulses in future. For those, who could not attend, please go to [www.hptlc.com](http://www.hptlc.com) for some meeting impressions and hopeful enjoyment. By the way the first CBS article of this issue was impressively presented at the symposium. Nice to hear that in the booming health care business, HPTLC analysis is again being used.

Sincerely,

*Gerda Morlock*

Gerda Morlock  
cbs@camag.com

# CAMAG

MÄRZ  
MARCH  
2007

# 98

# THE CBS CLASSIFICATION SYSTEM

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1. **Reviews and books**
  - a) Books on TLC
  - b) Books containing one or several chapters on TLC
  - c) Books containing frequent TLC information spread over several chapters of other information
2. **Fundamentals, theory and general**
  - a) General b) Thermodynamics and theoretical relationship
  - c) Relationship between structure and chrom. behaviour
  - d) Measurement of physico-chemical and related values
  - e) Optimization of solvent systems
  - f) Validation of methods
3. **General techniques** (unless they are restricted to the application within one or two classification sections)
  - a) New apparatus/techniques for sample preparation
  - b) Separation material
  - c) New apparatus for sample application/dosage
  - d) New apparatus/techniques for chromatogram development
  - e) New apparatus/techniques for pre- or post-chromatographic derivatization
  - f) New apparatus/techniques for quantitative evaluation
  - g) New apparatus/techniques for other TLC steps (distinguished from section 4)
4. **Special techniques**
  - a) Automation of sample preparation/application
  - b) Automation of complex chromatogram developing techniques
  - c) Automation, computer application in quantitative chromatogram evaluation
  - d) Combination of TLC with other chromatographic techniques
  - e) Combination of TLC with other (non-chromatographic) techniques...MS, IR...etc.
5. **Hydrocarbons and halogen derivatives**
  - a) Aliphatic hydrocarbons
  - b) Cyclic hydrocarbons
  - c) Halogen derivatives
  - d) Complex hydrocarbon mixtures
6. **Alcohols**
7. **Phenols**
8. **Substances containing heterocyclic oxygen**
  - a) Flavonoids
  - b) Other compounds with heterocyclic oxygen
9. **Oxo compounds, ethers and epoxides**
10. **Carbohydrates**
  - a) Mono- and oligosaccharides, structural studies
  - b) Polysaccharides, mucopolysaccharides, lipopolysaccharides
11. **Organic acids and lipids**
  - a) Organic acids and simple esters
  - b) Prostaglandins
  - c) Lipids and their constituents
  - d) Lipoproteins and their constituents
  - e) Glycosphingolipids (gangliosides, sulfatides, neutral glycosphingolipids)
12. **Organic peroxides**
13. **Steroids**
  - a) Pregnane and androstane derivatives
  - b) Estrogens
  - c) Sterols
  - d) Bile acids and alcohols
  - e) Ecdysones and other insect steroid hormones
14. **Steroid glycosides, saponins and other terpenoid glycosides**
15. **Terpenes and other volatile plant ingredients**
  - a) Terpenes
  - b) Essential oils
16. **Nitro and nitroso compounds**
17. **Amines, amides and related nitrogen compounds**
  - a) Amines and polyamines
  - b) Catecholamines and their metabolites
  - c) Amino derivatives and amides (excluding peptides)
18. **Amino acids and peptides, chemical structure of proteins**
  - a) Amino acids and their derivatives
  - b) Peptides and peptidic proteinous hormones
19. **Proteins**
20. **Enzymes**
21. **Purines, pyrimidines, nucleic acids and their constituents**
  - a) Purines, pyrimidines, nucleosides, nucleotides
  - b) Nucleic acids, RNA, DNA
22. **Alkaloids**
23. **Other substances containing heterocyclic nitrogen**
  - a) Porphyrins and other pyrroles
  - b) Bile pigments
  - c) Indole derivatives
  - d) Pyridine derivatives
  - e) other N-heterocyclic compounds
24. **Organic sulfur compounds**
25. **Organic phosphorus compounds** (other than phospholipids)
26. **Organometallic and related compounds**
  - a) Organometallic compounds
  - b) Boranes, silanes and related non-metallic compounds
  - c) Coordination compounds
27. **Vitamins and various growth regulators** (non-peptidic)
28. **Antibiotics, Mycotoxins**
  - a) Antibiotics
  - b) Aflatoxins and other mycotoxins
29. **Pesticides and other agrochemicals**
  - a) Chlorinated insecticides
  - b) Phosphorus insecticides
  - c) Carbamates
  - d) Herbicides
  - e) Fungicides
  - f) Other types of pesticides and various agrochemicals
30. **Synthetic and natural dyes**
  - a) Synthetic dyes
  - b) Chloroplasts and other natural pigments
31. **Plastics and their intermediates**
32. **Pharmaceutical and biomedical applications**
  - a) Synthetic drugs
  - b) Pharmacokinetic studies
  - c) Drug monitoring
  - d) Toxicological applications
  - e) Plant extracts
  - f) Clinico-chemical applications and profiling body fluids
  - g) Herbal and traditional medicines
33. **Inorganic substances**
  - a) Cations
  - b) Anions
34. **Radioactive and other isotopic compounds**
35. **Other technical products and complex mixtures**
  - a) Surfactants
  - b) Antioxidants and preservatives
  - c) Various specific technical products
  - d) Complex mixtures and non-identified compounds
36. **Thin-layer electrophoresis**
37. **Environmental analysis**
  - a) General papers
  - b) Air pollution
  - c) Water pollution
  - d) Soil pollution
38. **Chiral separations**

XX. (abstract number underlined) refers to HPTLC related publication or application using HPTLC materials

## 1. Reviews and books

- 98 004 J. SHERMA (Dept. of Chem., Lafayette Col., Easton, Pennsylvania 18042): Planar chromatography. *Anal. Chem.* 74, 2653-2662 (2002). This reviews covers the literature of TLC and HPTLC found by computer-assisted searching in Chemical Abstracts and the ICI Web of Science from November 1, 1999 to November 1, 2001. The literature search was augmented by consulting Analytical Abstracts, Chemical Titles and Current Contents, and the following journals were searched directly: *J. Chrom.* (parts A and B and the bibliography issues), *J. Chrom. Science*, *Chromatographia*, *Anal. Chem.*, *J. Liq. Chrom. & Rel. Technol.*, *J. AOAC Int.*, *J. Planar Chromatogr. - Modern TLC and Acta Chrom.* Publications in the past 2 years on the history, theory, methodology, instrumentation, and applications of TLC are discussed. 198 references are listed.

review

1

- 98 005 J. SHERMA (Dept. of Chem., Lafayette Col., Easton, Pennsylvania 18042): Planar chromatography. *Anal. Chem.* 72, 9R-25R (2000). Selective review of the literature of TLC and HPTLC in Chemical Abstracts from November 1, 1997 to November 1, 1999. The literature search was augmented by consulting Analytical Abstracts, Chemical Titles and Current Contents, and the following journals were searched directly: *J. Chrom.* (parts A and B and the bibliography issues), *J. Chrom. Science*, *Chromatographia*, *Anal. Chem.*, *J. Liq. Chrom. & Rel. Technol.*, *J. AOAC Int.*, *J. Planar Chromatogr. - Modern TLC and Acta Chrom.* The different chapters are: history, student experiments, books and reviews; theory and fundamental studies; chromatographic systems (stationary and mobile phases); apparatus and techniques; detection and identification of separated zones; quantitative analysis; preparative layer chromatography, thin-layer radio-chromatography and applications of TLC/HPTLC. 415 references from the last 2 years.

review

1

- 98 006 J. SHERMA (Dept. of Chem., Lafayette College, Easton, Pennsylvania 18042): Planar chromatography. *Anal. Chem.* 78, 3841-3852 (2006). TLC and HPTLC literature found by computer-assisted searching in Chemical Abstracts and the ISI Web of Science (11.2003 - 11.2005) is covered. The literature search was augmented by consulting Analytical Abstracts, and the following journals were searched directly: *J. Chrom.* (parts A and B and the bibliography issues), *J. Chrom. Science*, *Chromatographia*, *Anal. Chem.*, *J. Liq. Chrom. & Rel. Technol.*, *J. AOAC Int.*, *J. Planar Chromatogr. - Modern TLC and Acta Chrom.* (Papers reporting research on paper chromatography were not included.) 205 references were cited from the last 2 years. The different chapters are: history, student experiments, books and reviews; theory and fundamental studies; chromatographic systems (stationary and mobile phases); apparatus and techniques; detection and identification of separated zones; quantitative analysis; preparative layer chromatography and thin-layer radiochromatography.

review

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## 2. Fundamentals, theory and general

- 98 007 J. GILBERT\*, E. ANKLAM (\*Department for Environment, Food and Rural Affairs, Central Science Laboratory, Sand Hutton, York, UK, j.gilbert@csl.gov.uk): Validation of analytical methods for determining mycotoxins in foodstuffs. *TrAC* 21, 468-486 (2002). The article describes the valuable lessons learned from EU while funding a method-validation project (1996-2000) to meet European mycotoxin control in foodstuffs. It shows the performance characteristics of validated and official methods for aflatoxins, the selection and development of methods for validation, and the preparation of naturally contaminated mycotoxin test materials for validation studies. The authors put special emphasis on validation of TLC methods for mycotoxins for developing countries, as the main exporters to Europe of food and food products.

food analysis, toxicology, HPTLC, review

2f

- 98 008 K. FODOR\*, Z. VEGH, B. RENGER (\*Gedeon Richter Ltd., P.O.B. 27, H-1475 Budapest, Hungary, k.fodor@richter.hu): Thin-layer chromatography in testing the purity of pharmaceuticals. *TrAC* 25, 8 (2006). In the individual monographs of drug substances or finished products, only semi-quantitative TLC purity tests are mentioned and the number of TLC applications is steadily decreasing, being replaced by HPLC methods that are considered more appropriate. However, to comply with the latest and current pharmaceutical regulations, TLC manufacturers do not stop developing new equipment and accessories related to sample application, developing chambers, derivatization, documentation, and quantitative evaluation. Numerous examples of TLC/HPTLC applications in analytical research and quality control are mentioned to show the validity of this technique in the description of organic related impurities in drug substances and final products. Finally, authors ask analysts to present excellent, fully-validated and documented GMP/GLP-compliant TLC purity-test procedures to convince experts from pharmacopoeial committees and regulatory bodies of the importance of this analytical tool.
- quality control, comparison of methods 2a
- 98 009 J. K. ROZYLO\*, M. JANICKA, R. SIEMBIDA (\*Fac. of Chem., Maria Curie-Sklodowska Univ., 20031 Lublin, Poland): Different liquid chromatographic techniques in the study of the process of adsorption chromatography. *Acta Chrom.* 6, 21-38 (1996). The possibility of using a thermodynamic approach for optimization of mixed mobile phases for systems with binary and ternary solvents is discussed, as well as the method for predicting separation results in column liquid chromatography on the basis of experimental thin-layer chromatographic data. The effect of mobile phase composition on solute retention is described by solute capacity factors (characteristic of the pure solvents), molecular interactions occurring in the bulk phase and adsorption equilibrium in a given adsorbent - binary solution system. Two kinds of chambers were used in the investigation: a saturated Stahl chamber and a sandwich chamber. The study described the use of different TLC techniques as pilot methods for HPLC.
- binary/ternary mobile phase, optimization of mobile phase composition 2a
- 98 010 W. PRUS (Fac. of Textile Engineering and Environmental Protection, Univ. of Bielsko-Biala, 2 Wilowa Street, Bielsko-Biala, Poland): Relationship between raman scattering intensity in the aromatic region and the thermally induced increase in the UV absorption of chemically bonded stationary phases. *Acta Chrom.* 16, 204-215 (2006). Discussion of the relationship between the thermally induced increase in UV absorption of RP-8 and RP-18 layers and Raman-scattering intensity in the aromatic region when the stationary phases are irradiated with a high-power neodymium laser. The results obtained confirm the assumption of a linear relationship between the growth of UV absorption in this region by the stationary phase as a result of its thermal modification and the density of coverage of the silica gel with alkyl (octyl or octadecyl) ligands capable of aromatization.
- UV absorption, Raman scattering 2a
- 98 011 C. SARBU\*, B. TIPERCIUC (\*Babes-Bolyai University, Faculty of Chemistry and Chemical Engineering, Arany Janos 11, 400028 Cluj-Napoca, Romania): Modeling, by multivariate regression methods, of the chromatographic retention (lipophilicity) of new oxadiazoline derivatives. *J. Planar Chromatogr.* 19, 342-347 (2006). HPTLC of 13 oxadiazoline derivatives on RP-18 with methanol - water mixtures containing from 25 to 75 % methanol in steps of 10 %. After development the dried plates were examined under UV light at 254 nm. The retention indices predicted were satisfactory and in very good agreement with the molecular structure of the compounds investigated.
- HPTLC 2c

- 98 012 B. SPANGENBERG (University of Applied Sciences Offenburg, Badstrasse 24, 77625 Offenburg, Germany): Does the Kubelka-Munk theory describe TLC evaluations correctly? *J. Planar Chromatogr.* 19, 332-341 (2006). In TLC the development step distributes the sample throughout the layer. The essential requirement for quantitative TLC is a constant sample distribution in each sample spot. The paper shows that quantitative TLC is possible even if the concentration of the sample is not constant. In the absence of uniform sample distribution classical Kubelka-Munk theory must be extended. The extended theory presented is not only capable of describing asymmetrical scattering in TLC layers but also includes a formula for absorption and fluorescence in diode-array TLC. With this new formula all different formulas for diode-array thin-layer chromatographic evaluation are combined in one expression.

2b

### 3. General techniques

- 98 013 C. B. FANG (Congbing Fang), X. C. WAN\* (Xiaochun Wan), C. J. JIANG (Changjun Jiang), H. R. TAN (Huarong Tan), Y. H. HU (Yinghui Hu), H. Q. CAO (Haiqun Cao) (\*Key Laboratory of Tea Biochemistry and Biotechnology, Ministry of Education and Ministry of Agriculture, Anhui Agricultural University, Hefei 230036, China): Comparison of HPTLC, HPLC, and HPCE for fingerprinting of *Pueraria Radix*. *J. Planar Chromatogr.* 19, 348-354 (2006). HPTLC of puerarin, daidzein, daidzin, and 3'-methoxypuerarin on silica gel, pre-washed with methanol, in an unsaturated twin-trough chamber with chloroform - methanol - ethyl acetate - water 16.2:18.8:52:3. Quantitative determination by absorbance measurement at 254 nm. The relative standard deviation of R<sub>f</sub> values, retention times and peak area percentages all meet the national standards.

traditional medicine, herbal, HPTLC, quantitative analysis, densitometry 3e

- 98 014 G. GRYGIERCZYK (Inst. of Chem., Silesian Univ., 9 Szkolna Street, 40-006 Katowice, Poland): Chromatographic analysis of organic compounds on impregnated chemically bonded stationary phases. Part I. *Acta Chrom.* 17, 302-313 (2006). Non-polar (RP-2, RP-8, and RP-18) and polar (amino, cyano, and diol) chemically bonded stationary phases have been impregnated with solutions of organic substances at different concentrations and the effect of impregnation on the mechanism of retention of alcohols, higher fatty acids, amino acids, and medicines has been investigated.

comparison of methods, impregnation, retention behaviour 3b

- 98 015 T. HALKINA, J. SHERMA\* (\*Dept. of Chem., Lafayette Col., Easton, PA 18042-1782, USA): Comparative evaluation of the performance of silica gel TLC plates and irregular and spherical-particle HPTLC plates. *Acta Chrom.* 17, 261-271 (2006). TLC and HPTLC plates have been compared on the basis of theoretical plate number, resolution, linearity, development time, and limit of sensitivity for analysis of a multicomponent analgesic tablet in the fluorescence quenching mode and analysis of a five-component dye mixture in the visible mode.

HPTLC, comparison of methods, irregular and spherical HPTLC plates 3b

- 98 016 T. HALKINA, J. SHERMA\* (\*Dept. of Chem., Lafayette Col., Easton, PA 18042-1782, USA) : Use of the ChromImage flatbed scanner for quantification of high-performance thin layer chromatograms in the visible and fluorescence-quenching mode. *Acta Chrom.* 17, 250-260 (2006). The ChromImage flatbed scanner densitometer with Galaxie-TLC software has been used for the quantification of silica gel HPTLC. The visible mode was evaluated by determination of the recovery of rhodamine B from a four-dye mixture and by determination of the precision of replicate analysis. Determination of caffeine in analgesic tablets and a cola beverage were performed in the fluorescence-quenching mode.

- HPTLC, quantitative analysis, flatbed scanner 3f
- 98 017 A. N. CAMPBELL\*, J. SHERMA (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA): Comparative evaluation of precoated silica gel plates for preparative layer chromatography. *Acta Chrom.* 13, 102-108 (2003). Five commercial preparative layer chromatography plates precoated with silica gel of 1 mm thickness were compared on the basis of theoretical plate number and resolution by developing a test dye mixture (1.0 - 5.0 µg per zone) with ethyl acetate - methanol - water 4:1:1. Best results were obtained with the Mallinckrodt-Baker layer with 4.5 - 5.5 µm spherical particles. With one exception the efficiency and resolution of the other layers correlated with their particle size.  
preparative TLC, comparison of methods, stationary phases 3b
- 98 018 Nada PERISIC-JANJIC\*, T. DJAKOVIC-SEKULIC (\*Department of Chemistry, Faculty of Science, University of Novi Sad, Trg D. Obradovica 3, 21000 Novi Sad, Serbia): Study of the characteristics and separation properties of unconventional TLC supports. Part I. *J. Planar Chromatogr.* 19, 438-442 (2006). TLC of linear aliphatic alcohols with 1 to 20 carbon atoms, as the esters of 3,5-dinitrobenzoic acid, on silica gel impregnated with paraffin oil, aminoplast, microcrystalline cellulose, rice starch, and talc with water - ethyl methyl ketone - dioxane (35:8:X where X = 15-70). Detection by spraying with 0.01 % solution of rhodamine B in methanol, followed by evaluation under UV light at 366 nm. The stationary phases were characterized by use of separation factor, resolution, and delta RF values.  
qualitative identification 3b
- 98 019 W. PRUS (Faculty of Textile Engineering and Environmental Protection, University of Bielsko-Biala, Bielsko-Biala, Poland): GC-MS study of the products cleaved from RP-18 chemically bonded phases for TLC during thermal modification. *J. Planar Chromatogr.* 19, 324-326 (2006). GC-MS investigation of dichloromethane extracts from three different types of RP-18 modified silica gel used as TLC adsorbents. Alkenes and carbonyl compounds (two different aldehydes and one ketone) were identified.  
3b
- 98 020 A. V. GERASIMOV, (All-Russian Scientific Institute of Food Flavorings, Acids, and Dyes (GU VNIIPAKK), Liteinyi pr. 55, St. Petersburg, 191104, Russia): Use of the software processing of scanned chromatogram images in quantitative planar chromatography. *J. Anal. Chem.* 59 (4), 348-353 (2004). Demonstration of quantitative analysis using the software processing of scanned chromatogram images for e.g. food dyes. Digitalization of chromatograms obtained by scanning with a flatbed scanner using the special-purpose software for quantitative analysis.  
quantitative analysis, software processing, scanning 3f
- 98 021 J. WANG (Jie Wang), D. WANG\* (Dongyuan Wang), H. ZHANG (Hongxia Zhang), Y. ZHANG (Yanhua Zhang), S. ZHOU (Shuyu Zhou) (\*College of Pharmacy, Shenyang Pharmaceutical University, Shenyang 110016, P.R. China): A new plate for planar electrochromatography. *J. Planar Chromatogr.* 19, 313-318 (2006). A new bonded RP-18 sintered silica gel plate for use in reversed-phase planar electrochromatography (PEC) has been prepared by fusing a mixture of silica gel and glass powder then reaction with octadecyltrichlorosilane. The plates have good chromatographic characteristics and good mechanical stability, and can be regenerated by soaking in acetone. TLC and PEC of p-aminoazobenzene, azobenzene, p-hydroxyazobenzene, Sudan II, and Sudan III with acetonitrile - water 17:3.  
electrochromatography 3b

#### 4. Special techniques

- 98 022 A. CRECELIUS\*, M. R. CLENCH, D. S. RICHARDS (\*Biomed. Res. Centre, Sheffield Hallam Univ., Sheffield, UK): TLC-MALDI in pharmaceutical analysis. *LC-GC Europe* 16, Issue 4, 225-229 (2003). A technique for the direct determination of substances from TLC plates by MALDI-MS is discussed. Methods for the generation of quantitative data by adding an internal standard to the development solvent are described and the use of post-source decay-MALDI experiments in conjunction with TLC-MALDI-MS for compound identification is reported.  
pharmaceutical research, online TLC-MS 4e
- 98 023 M. J. FORD, M. A. DEIBEL, B. A. TOMKINS, G. J. VAN BERKEL\* (\*Org. and Biol. Mass Spectrometry Group, Chem. Science Division, Oak Ridge Nat. Lab., Oak Ridge, Tennessee 37831-6131): Quantitative thin-layer chromatography/mass spectrometry analysis of caffeine using a surface sampling probe electrospray ionization tandem mass spectroscopy system. *Anal. Chem.* 77, 4385-4389 (2005). TLC of caffeine on RP-8 with methanol - water (7:3) followed by heating at 110° C for 15 min. Detection at 214 nm. Quantification of caffeine at the low-nanogram level is performed directly from a surface of the RP-8 TLC plate using a surface sampling probe and an ESI mass spectrometry system employing selected reaction monitoring detection, and a deuterium-labelled internal standard spotted with the samples. TLC/ESI-MS/MS successfully allowed caffeine quantification in six commercially available beverages using only minimal sample preparation. The resulting calculated analyte concentrations exhibited accuracy comparable to the HPLC/UV method. However, the surface sampling probe used was restricted to non-polar layers.  
quantitative analysis, qualitative identification, online TLC-MS, caffeine 4e
- 98 022 G. J. VAN BERKEL\*, A. D. SANCHEZ, J. M. E. QUIRKE (\*Org. and Biol. MS Group, Chem. Sc. Div. Oak Ridge Nat. Lab., Oak Ridge, Tennessee 37831-6131): Thin-layer chromatography and electrospray mass spectrometry coupled using a surface sampling probe. *Anal. Chem.* 74, 6216-6223 (2002). A combined surface sampling probe/electrospray emitter was used for the direct readout of TLC plates by electrospray MS. HPTLC of methylene blue, crystal violet, and rhodamine 6G on RP-18 with methanol - tetrahydrofuran 3:2 containing 50-100 mM ammonium acetate, followed by MS detection in positive ion mode. HPTLC of fluorescein, naphtholblue black, and fast green FCF on PR-18 with methanol - water 7:3, followed by MS detection in negative ion mode. Acquisition of mass spectra of components of individual bands was shown by manual stepping to and sampling from specific locations within the bands. Computer-controlled scanning of lanes was illustrated by using multiple ion monitoring in positive and negative ion modes. Readout resolution, the limits of scan speed, detection levels, TLC phase (restricted to nonpolar phases), and eluting solvents were investigated.  
HPTLC, online TLC-MS, surface sampling 4e
- 98 023 G. J. VAN BERKEL\*, V. KERTESZ (\*Org. and Biol. Mass Spectrometry Group, Chem. Science Division, Oak Ridge Nat. Lab., Oak Ridge, Tennessee 37831-6131): Automated sampling and imaging of analytes separated on thin-layer chromatography plates using desorption electrospray ionization mass spectrometry. *Anal. Chem.* 78, 4938-4934 (see correction p.6283) (2006). Modest modifications to the atmospheric sampling capillary of a commercial electrospray mass spectrometer are described. Discussions of upgrades to a developed surface positioning control software package used to enable the automated sampling and imaging of analytes on and within large area surface substrates using desorption electrospray ionization mass spectrometry. Application of rhodamine B, 6G and 123 on TLC RP-8 plates with methanol - water containing 500 mM ammonium acetate 3:1. Examples are shown for user-defined spot sampling from separated bands on a TLC plate, scanning of the complete development lane, or imaging of analyte bands in

a development lane.

online TLC-MS

4e

- 98 024 M. LANCASTER, D. M. GOODALL\*, E. T. BERGSTROEM, S. MCCROSSEN, P. MYERS (\*Dept. of Chem., Univ. of York, York YO10 5DD, UK): Real-time image acquisition for absorbance detection and quantification in thin-layer chromatography. *Anal. Chem.* 78, 905-911 (2006). First quantitative study of real-time image acquisition of TLC plates during development. Procedures are described for imaging using a CCD camera and for image processing, incorporating corrections for fixed pattern effects and compensation for the moving solvent front, to measure the absorbance of the analyte. Example of use was the TLC and HPTLC separation of Sudan II on silica gel with dichlormethane in a horizontal chamber. The integrated peak areas were found to be independent of time and distance moved. The method gives limits of detection better than those from offline measurements on wetted TLC plates. In addition, obtained information on the degree of solvent permeation into the stationary phase at the analyte position could be useful for TLC and HPLC method development.

quantitative analysis, real-time image acquisition

4e

- 98 025 I. MALINOWSKA (Dept. of Adsorption and Planar Chrom., Fac. of Chem., Maria Curie Skłodowska Univ., Pl. Marii Curie - Skłodowskiej 3, 20-031 Lublin, Poland): Some aspects of the effect of an electric field in reversed-phase thin-layer chromatography. *Acta Chrom.* 11, 204-214 (2001). Electric fields have been shown to affect peak width, peak area, and other chromatographic properties in reversed-phase TLC on silanized silica gel and RP-18. To eliminate electroosmotic flow, thin layer electro-chromatography was performed on unwetted layers. Horizontal and vertical modes of thin-layer electrochromatography were investigated, but only for chromatographic systems in which van der Waals forces are the sole interactions. Analytes were polycyclic aromatic hydrocarbons and hexane or heptane was used as mobile phase. The electric field affects the chromatographic process not only by replacing capillary flow by electroosmotic flow.

electrochromatography, electroosmotic flow

4d

- 98 026 I. MEISEN, A. FRIEDRICH, H. KARCH, U. WITTING, J. PETER-KATALINIC, J. MÜTHING\* (\*Institute for Medical Physics and Biophysics, University of Munster, Münster, Germany, jm@uni-muenster.de): Application of combined high-performance thin layer chromatography immunostaining and nanoelectrospray ionization quadrupole time-of-flight tandem mass spectrometry to the structural characterization of high- and low-affinity binding ligands of Shiga toxin 1. *Rapid Commun. Mass Sp.* 19, 3659-3655 (2005). HPTLC of a preparation of neutral Stx1-binding glycosphingolipids from human erythrocytes, comprising 21.4 % and 59.1 % of the high- and low-affinity Stx1-binding ligands Gb3Cer/CD77 and Gb4Cer respectively and their antibody positive bands on silica gel with chloroform - methanol - water 15:30:4. Crude extracts were used without any further purification for an analysis by nanoelectrospray ionization quadrupole time-of-flight mass spectrometry in the negative mode. Only analytical quantities in the microgram scale of a single glycosphingolipid species are required for the structural MS characterization.

clinical chemistry research, HPTLC, preparative TLC

4e

- 98 027 K. NAKAMURA, Y. SUZUKI, N. GOTO-INOE, C. YOSHIDA-NORO, A. SUZUKI\* (\*Sphingolipid Expression Lab., Supra-Biomolecular System Research Group, Frontier Research System, Inst. of Phys. and Chem. Research (RIKEN), Saitama, Japan): Structural characterization of neutral glycosphingolipids by thin-layer chromatography coupled to matrix-assisted laser desorption/ionization quadrupole ion trap time-of-flight MS/MS. *Anal. Chem.* 78, 5736-5743 (2006). Structural analysis of neutral glycosphingolipids by direct coupling of TLC to MALDI-QIT-TOF MS/



MS. TLC of neutral glycosphingolipids on silica gel with chloroform - methanol - water 65:35:8. The underivatized glycosphingolipid spots were directly subjected to MALDI-QIT-TOF MS after addition of a matrix solution of 2,5-dihydrobenzoic acid in acetonitrile - water 1:1, which proved to be suitable for MS/MS analysis with high sensitivity. MS/MS and MS/MS/MS spectra reveal the ceramide and long-chain base structures, as well as the sugar sequences. Furthermore, derivatization with primuline (0,01 % in acetone - water, 4:1), a non-destructive fluorochrome for lipid detection, was used to locate glycosphingolipids on a TLC plate prior the MS analysis. The coupling of TLC-immunostaining of glycosphingolipids to MALDI-QIT-TOF MS/MS is shown to be a powerful method to identify both the antibody-specific sugar and the ceramide structures.

online TLC-MS, glycosphingolipids, immunostaining

4e

- 98 028 S. NYIREDI (Res. Inst. for Med. Plants, Budakalász, Hungary): Multidimensional Planar Chromatography. LC-GC Europe 16, Issue 12a, 52-59 (2003) LC-GC Europe 16, 52-59 (2003). The potential of various multidimensional planar chromatography (MD-PC) techniques, including comprehensive 2D planar chromatography (PC×PC), targeted or selective 2D PC (PC+PC), modulated 2D PC (nPC) and coupled-layer PC (PC-PC) is discussed.

multidimensional planar chromatography

4d

- 98 029 A. PARENT, T. ANDERSON, D. MICHAELIS, G. JIANG, P. SAVAGE, M. LINDFORD\* (\*Department of Chemistry and Biochemistry, Brigham Young University, Provo, UT 84602, United States, mrlindford@chem.byu.edu): Direct ToF-SIMS analysis of organic halides and amines on TLC plates. Appl. Surf. Sci. 252, 6746-6749 (2006). To show the application of direct time-of-flight secondary ion mass spectrometry (ToF-SIMS) to analyze a reaction mixture between an alpha-bromoamide (1), and picolylamine, to form the substitution product (3) via SN2 (1 and 3 are indistinguishable by TLC on silica gel with dichloromethane - methanol - ammonia 100:15:2), a series of organic halides with widely differing structures (trifluoromethanesulfonamide, a vinyl bromide, a silyl-protected primary bromide, and iodosobenzene diacetate) were separated on TLC plates under the same conditions. The resulting spots and background were analyzed by positive and negative ion spectra. In all cases, the halide signals were notably stronger than the background signals. In addition, a series of amines (1,1-carbonyl diimidazole, norepinephrine, adenosine, and cinchonidine) was separated on TLC plates and analyzed directly by ToF-SIMS. In all but one (adenosine), even quasi-molecular ions were observed.

pharmaceutical research, qualitative identification

4e

## 6. Alcohols

- 98 030 M. WAKSMUNDZKA HAJNOS\*, G. JOZWIAK, T. WAWRZYNOWICZ (\*Dept. of Inorg. and Anal. Chem., Med. Univ., Staszica 6, 20-081 Lublin, Poland): RP-TLC and RP-HPLC analysis of aliphatic alcohols as the DANS-Cl derivatives. Acta Chrom. 11, 51-61 (2001). A homologous series of C4 - C12 aliphatic alcohols was derivatized with dansyl chloride to introduce a fluorophoric group into the molecule which enables TLC and/or HPLC analysis with UV detection. The progress of the reaction under different sets of conditions was monitored by RP-TLC and it was confirmed that the best yield was obtained after 24 h at ambient temperature. TLC on RP-18 with acetonitrile - water 3:2 in a horizontal chamber. Densitometry at 361 nm.

qualitative identification, densitometry, aliphatic alcohols, dansyl chloride

6

## 7. Phenols

- 98 031 U. HUBICKA, J. KRZEK\*, J. KALETA, A. NIEDZWIEDZ (\*Jagiellonian University, Collegium Medicum, Department of Inorganic and Analytical Chemistry, Medyczna 9, 30-688 Cracow,

Poland): Evaluation of densitometric TLC for quantitative analysis of selected phenolic acids for standardization of propolis concentrates. *J. Planar Chromatogr.* 19, 449-453 (2006). HPTLC of caffeic, p-coumaric, and ferulic acid on silica gel with dichloromethane - acetonitrile - 90 % formic acid 95:5:1. Quantitative determination by absorbance measurement at 320 nm. The method is precise and accurate.

food analysis, quality control, HPTLC, densitometry, quantitative analysis 7

98 122 A. MOHAMMAD et al., see section 33

98 032 A. MORNAR, Marica MEDIC-SARIC\*, I. JASPRICA (\*Department of Medicinal Chemistry, University of Zagreb, A. Kovacica 1, 10 000 Zagreb, Croatia): ADME data for polyphenols characterized by reversed-phase thin-layer chromatography. *J. Planar Chromatogr.* 19, 409-417 (2006). The chromatographic behavior of polyphenols (flavonoids and phenolic acids) has been investigated by RP-TLC to establish relationships between chromatographic data and selected ADME (absorption, distribution, metabolism, and elimination) data. TLC of six flavones (flavone, chrysin, tectochrysin, apigenin, acacetin, luteolin), nine flavonols (galangin, kaempferol, kaempferide, morin, quercetin, rhamnetin, isorhamnetin, tamarixetin, myricetin), seven flavanones (flavanone, pinocembrine, pinostrobin, naringenin, sakuranetin, isosakuranetin, hesperitin), and eight phenolic acids (caffeic, ferulic, isoferulic, cinnamic, o-coumaric, m-coumaric, p-coumaric, and sinapic acid) on RP-18. Binary mobile phases were prepared from mixtures of methanol and phosphate buffer; ascending development was performed at room temperature in a saturated glass chamber. Detection at 254 nm.

pharmaceutical research, herbal, qualitative identification 7

## 8. Substances containing heterocyclic oxygen

98 033 P. K. SALO, A. ESSÉN-SUURONEN, H. SALOMIES, R. A. KETOLA, R. KOSTIAINEN\* (\*Faculty of Pharmacy, Division of Pharmaceutical Chemistry, P. O. Box 56, FIN-00014 University of Helsinki, Finland): HPTLC, with UV and MS detection, and preparative-layer chromatography for analysis and purification of synthesis products. *J. Planar Chromatogr.* 19, 371-377 (2006). HPTLC of five isoflavone products and 2-phenylchromone as reference standard on silica gel, pre-washed with acetonitrile, in an unsaturated chamber, twice with chloroform or dichloromethane. Determination by absorbance measurement at 280 nm. Preparative layer chromatography on silica gel. A new device for isolation of synthesis products in sub-milligram amounts was successfully employed.

qualitative identification, preparative TLC, HPTLC, synthetic organic chemistry 8a

98 034 Agnieszka SKALSKA\*, A. MATYSIK, M. GERKOWICZ, M. WÓJCIAK-KOSIOR (\*Department of Chemistry, Laboratory of Planar Chromatography, Medical Academy, Staszica 6, 20-081 Lublin, Poland): Preparative reversed-phase high-performance thin-layer chromatography for analysis of anthocyanins. *J. Planar Chromatogr.* 19, 463-466 (2006). Preparative RP-HPTLC of anthocyanin extracts (e. g. malvidine) on silica gel with five-step gradient elution with different concentrations of toluene - acetonitrile - water - formic acid - n-butanol - 2-propanol and resp. addition of tert-butylmethyl ether. After extraction of the third zone from the layer, HPTLC on RP-18 as the best stationary phase with a three-step gradient elution using methanol - water - formic acid in different ratios. Quantitative determination by absorbance measurement at 470 nm.

herbal, quality control, preparative TLC, HPTLC, densitometry, quantitative analysis, qualitative identification 8a

## 10. Carbohydrates

- 98 035 S. D. WAGNER\*, J. PACHUSKI, B. FRIED, J. SHERMA (\*Dept. of Chem., Lafayette Col., Easton, Pennsylvania 18042, USA): Thin layer chromatographic analysis of carbohydrates and amino acids in *Schistosoma mansoni* (Trematoda) cercariae. *Acta Chrom.* 12, 159-169 (2002). TLC of carbohydrates on LK5DF silica gel plates containing 19 lanes and a pre-adsorbent sample-application zone, prewashed with dichloromethane - methanol 1:1, with ethyl acetate - glacial acetic acid - methanol - water 12:3:3:2. Detection by spraying with alpha-naphthol-sulfuric acid sugar-detection reagent followed by heating at 110 °C for 5 min. Densitometry of glucose and raffinose at 515 nm. HPTLC of amino acids with four different TLC systems: on silica gel with preadsorbent zone, on RP-18 with concentrating zone, on cellulose F with n-butanol - acetic acid - water 3:1:1, and on ion-exchange sheets with citrate buffer of pH 3.3 (84 g citric acid monohydrate, 16 g NaOH, and 5.9 mL HCl 37% per L). Detection by spraying with ninhydrin reagent, followed by drying in air for 30 min and heating for 10 min at 110 °C. Quantification by densitometry at 495 nm for histidine and 610 nm for tryptophan.

pharmaceutical research, quantitative analysis, qualitative identification, HPTLC, carbohydrates, amino acids  
10a

- 98 036 D. J. CLINE, B. FRIED, J. SHERMA\* (\*Dept. of Chem., Lafayette College, Easton, PA 18042, USA): TLC and GC-MS identification of glucose and maltose in *Biomphalaria glabrata* (Gastropoda), and use of quantitative TLC to determinate the effect of starvation on the amounts of these carbohydrates. *Acta Chrom.* 9, 79-86 (1999). HPTLC of the primary carbohydrates in the planorbid snail *Biomphalaria glabrata* on silica gel with acetonitrile - water 17:3 and ethyl acetate - acetic acid - methanol - water 12:3:3:2, on LK5DF silica gel with ethyl acetate - acetic acid - methanol - water 12:3:3:2, on RP-18 with tetrahydrofuran - water 22:3, on cellulose with ethyl acetate - pyridine - water 2:1:2 and on amino-bonded layers with ethyl acetate - pyridine - water - acetic acid 12:6:2:1. The use of specific detection reagents for reducing sugars (4-aminobenzoic acid detection reagent and aniline - DPA reagent), spiking experiments, and spraying with alpha-naphthol - sulfuric acid reagent for quantitative analysis (densitometry at 515 nm) aided the identification of glucose and maltose. GC-MS analysis confirmed the identification of maltose and glucose on the basis of retention times and spectral fingerprints. A starvation study was conducted to determine changes in sugar levels in *B. glabrata* digestive gland-gonad complex (DGG) and hemolymph samples during a 12-day starvation period.

pharmaceutical research, quantitative analysis, qualitative identification, HPTLC, densitometry, glucose, maltose  
10a

- 98 037 J. J. SCHARITER\*, B. FRIED, J. SHERMA (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA):: TLC analysis of glucose and maltose in *Biomphalaria glabrata* (Gastropoda) infected with *Schistosoma mansoni* (Trematoda). *Acta Chrom.* 11, 102-107 (2001). The snail *Biomphalaria glabrata* is medically important because it serves as the intermediate host for the development and transmission of the human blood fluke parasite *Schistosoma mansoni*. The purpose of this study was to use HPTLC to determine the effects of *S. mansoni* larval infection on the maltose and glucose content of the hemolymph and digestive gland-gonad complex of *B. glabrata*. TLC on silica gel with pre-adsorbent zone and 19 channels, with ethyl acetate - acetic acid - methanol - water 13:3:3:2. Detection by spraying with 1-naphthol - sulfuric acid reagent, quantitative determination by absorbance measurement at 515 nm.

pharmaceutical research, quantitative analysis, qualitative identification, densitometry, glucose, maltose  
10a

- 98 038 J. SHERMA\*, D. L. ZULICK (\*Dept. Of Chem., Lafayette Coll., Easton, PA 18042-1782, USA): Determination of fructose, glucose and sucrose in beverages by high-performance thin layer

chromatography. *Acta Chrom.* 6, 7-12 (1996). HPTLC of fructose, glucose and sucrose on silica gel with concentration zone. Impregnation of the layer (except concentration zone) by spraying with 0.1 M sodium bisulfate solution, followed by drying and spraying with citrate buffer of pH 4.8. Three-fold development with acetonitrile - deionized water 17:3 after chamber saturation. Visualization by spraying with 1-naphthol - sulfuric acid reagent and heating at 110 °C. Quantification by densitometry at 515 nm.

food analysis, quantitative analysis, densitometry, HPTLC, fructose, glucose, sucrose 10

- 98 039 A. SZABÓ, A. KÓNYA, I. WINKLER, G. MÁTÉ, B. ERDÉLYI\* (\*IVAX Drug Research Institute Ltd., P. O. Box 82, 1326 Budapest, Hungary): Simultaneous monitoring of target compounds and carbohydrate patterns during pharmaceutical fermentations. *J. Planar Chromatogr.* 19, 418-421 (2006). TLC of fructose, glucose, saccharose, maltose, maltotriose, maltotetraose, maltopentaose, maltohexaose on silica gel in a pre-saturated twin-trough chamber with chloroform - carbon tetrachloride - 35 % aqueous formic acid - methanol 20:5:17:22. After drying the chromatogram was immersed in a solution of sulfuric acid reagent (10 mL conc. sulfuric acid in 200 mL of a cooled 1:1 mixture of 1-propanol and carbon tetrachloride) and heated for 8 min at 108 °C. Quantitative determination by absorbance measurement at 290 nm.

qualitative identification, quantitative analysis, densitometry, pharmaceutical production 10

- 98 040 H. THERISOD, V. LABAS, M. CAROFF\* (\*Equipe "Endotoxines", UMR 8619 du Centre National de la Recherche Scientifique, Biochimie, Université de Paris-Sud, F-91405 Orsay, France): Direct microextraction and analysis of rough-type lipopolysaccharides by combined thin-layer chromatography and MALDI mass spectrometry. *Anal. Chem.* 73, 3804-3807 (2001). TLC of lipopolysaccharides from intact bacteria on silica gel with chloroform - methanol - water - triethylamine 12:6:1:0.04. Detection by spraying with 10 % sulphuric acid in ethanol followed by heating at 150 °C for 5 min. For the non-destructive visualization the plates were pre-washed twice with propanol - water 1:1 and developed in methanol - water 1:1 which made the products appear as dull white spots on a bright background. Zones containing lipopolysaccharides or lipid A molecular species were isolated from the plate and analyzed by MALDI mass spectrometry.

lipopolysaccharides

10b

## 11. Organic acids and lipids

- 98 041 F. M. HELMY (Dept. of Biol. Sci., Delaware State Univ., Dover, DE, USA): Cardiophilin, its preferential deacylation in mammalian myocardia. Mini review and chromatographic-computational analysis. *Acta Chrom.* 17, 9-19 (2006). Comparative correlative TLC analysis conducted on whole-tissue homogenate of control and in-vitro-incubated samples of mammalian myocardia and bullfrog cardiac muscle and thigh skeletal muscle. The focus in the study was on TLC-densitometric analysis of the endogenous cardiophilin of control samples and their respective incubated samples to identify the products, e. g. lysocardiophilin, formed during in-vitro incubation. TLC of the chloroform-methanol extracts (2:1) obtained after the incubation from the freeze-dried tissues homogenate on silica gel with 1-propanol - chloroform - ethyl acetate - methanol - water 50:50:50:21:18. Visualization with thionine reagent, densitometry at 600 nm.

quantitative analysis, densitometry, cardiophilin, monolysocardiophilin, phosphatidylethanolamine, plasmalogen

11c

- 98 042 M. WAKSMUNDZKA-HAJNOS\*, H. D. SMOLARZ, R. NOWAK (\*Dept. of Inorg. and Anal. Chem., Med. Acad., Staszica 6, 20-081 Lublin, Poland): Chromatographic separations of phenolic acids by normal-phase TLC. Retention behaviour on polar adsorbents (silica, alumina, polyamide) with non-aqueous mobile phases. *Acta Chrom.* 9, 38-54 (1999). TLC of benzoic and cinnamic acid derivatives on silica gel 60 H compared with that on alumina oxide 60 G and po-

lyamide layers with non-aqueous ternary mobile phases comprising a non-polar diluent (n-heptane), a polar modifier (2-propanol, dioxane, tetrahydrofuran and ethyl acetate) and 2 % acetic acid in horizontal chambers. Visualization under UV 254 nm, and by spraying with diazotized sulphanic acid in 10 % sodium bicarbonate solution, or with a 2 % solution of ferric chloride. Chromatogram parameters and the correlation diagrams were used for comparison of separation selectivity.

comparison of methods, qualitative identification, phenolic acids 11a

### 13. Steroids

98 043 M. BATHORI\*, H. KALASZ (\*Dept. of Pharmacognosy, Fac. of Pharm., Univ. of Szeged, Hungary): Separation methods for phytoecdysteroids. LC-GC Europe 14, Issue 10, 626-631 (2001). TLC of 20-hydroxyecdysone on silica gel, pre-washed with methanol. Different mobile phases are investigated, the mixture of chloroform - methanol - benzene 25:5:3 was used for quantification. Densitometry at 254 nm. TLC of ecdysteroids under the same conditions, detection at 254 nm, in white light and at 366/>400 nm (fluorescence) after spraying with a vanillin-sulphuric acid reagent. Identification was facilitated with two-dimensional TLC, displacement TLC (D-TLC) and forced flow TLC (FF-TLC). Comparison with HPLC.

pharmaceutical research, quantitative analysis, comparison of methods, ecdysteroids 13

98 044 B. JANOSKA\*, T. WIELKOSZYNSKI, K. TYRPIEN, C. DOBOSZ, D. BODZEK (\*Med. Univ. of Silesia, Fac. of Med., Dept. of Chem., Jordana 19, 41-808 Zabrze, Poland): Effect of steroid hormones on results from determination of oxysterol by TLC. Acta Chrom. 13, 95-101 (2003). Chromatographic systems comprising different stationary and mobile phases were investigated for determination of oxysterols in plasma by TLC. Two chromatographic systems are used to ensure selectivity. TLC on RP-18 with 2-propanol - dichloromethane 3:97, or on silica gel with acetone - chloroform 1:9 in a horizontal chamber. Detection by spraying with Liebermann - Burchard reagent (methanol - conc. H<sub>2</sub>SO<sub>4</sub> - acetic anhydride, 10:1:1) followed by heating at 110 °C for 5 min. Densitometry at 366 nm (oxysterols) and 254 nm (steroid hormones).

clinical chemistry research, pharmaceutical research, quantitative analysis, qualitative identification, oxysterols, steroid hormones 13c

### 14. Steroid glycosides, saponins and other terpenoid glycosides

98 045 H. AGRAWAL, N. KAUL, A. R. PARADKAR, K. R. MAHADIK\* (\*Dept. of Pharm. Anal. Chem., Bharati Vidyapeeth Deemed Univ., Poona Col. of Pharm., Erandwane, Pune-411038, Maharashtra State, India): Separation of bacoside A3 and bacopaside II, major triterpenoid saponins in *Bacopa monnieri*, by HPTLC and SFC. Application of SFC in implementation of uniform design for herbal drug standardization, with thermodynamic study. Acta Chrom. 17, 125-150 (2006). HPTLC of bacoside A3 and bacopaside II on RP-18 F254 after pre-washing with methanol and heating at 60° C for 5 min. Development with toluene - methanol - ethyl acetate 15:5:4 in the dark in a controlled humidity chamber (humidity of 55 - 65 %). Densitometry at 344 nm. The method is available for content determination of bacoside A3 and bacopaside II in herbal extracts and commercial formulations.

quantitative analysis, HPTLC, densitometry, saponins 14

### 17. Amines, amides and related nitrogen compounds

98 046 H. A. KHAN (Department of Biochemistry, College of Science, King Saud University, P.O. Box 2455, Riyadh, 11451 Saudi Arabia): TLC determination of aliphatic polyamines on calcium sulfate layers. Chromatographia 64 (7-8), 423-427 (2006). TLC of six polyamines (ornithine, cit-

rulline, putrescine, cadaverine, spermidine and spermine) on calcium sulfate (CaSO<sub>4</sub>) and silica gel with 11 different mobile phases, using methanol as the solvent to enhance selectivity and produce differential R<sub>f</sub> values. On CaSO<sub>4</sub> better separation was achieved than on silica gel, no derivatization is needed, and development time is about 1/3 shorter. Quantitative determination by absorbance measurement at 550 nm after extracting substances from the plates and eluting from the coating material. Limit of detection (LOD) was 750 ng, limit of quantification (LOQ) was 1880 ng. Quantitative separation of underivatized polyamines in spiked human urine samples.

pharmaceutical research, comparison of methods, calcium sulfate layer, separation of polar compounds, biogenic polyamines 17a

## 18. Amino acids and peptides, chemical structure of proteins

- 98 047 S. A. NABI\*, M. A. KHAN (\*Dept. of Chem., Aligarh Muslim Univ., Aligarh, India): Selective TLC separation of lysine and threonine in pharmaceutical preparations. *Acta Chrom.* 13, 161-171 (2003). Lysine and threonine were separated and quantitatively determined from the mixture of amino acids present in a commercially available drug. TLC of alpha-amino acids on layers prepared from a 1:4 stannic arsenate - cellulose mixture with a variety of mobile phases. Best separation was achieved with n-butanol - formic acid - water 7:2:1; or isopropanol - acetic acid - water 8:1:1, or isopropanol - formic acid - water 7:2:1. Detection by dipping in 1 % ninhydrin in butanol after heating at 60 °C. Quantitative determination by spectrophotometry with hydrindantin-methyl cellosolve reagent.

pharmaceutical research, quality control, quantitative analysis, qualitative identification, lysine, threonine 18a

- 98 048 F. BUHL, Monika GALKOWSKA\* (\*Institute of Chemistry, Department of Analytical Chemistry, Silesian University, 9 Szkolna Street, 40-006 Katowice, Poland): Determination of methionine in pharmaceuticals after chromatographic separation. *J. Planar Chromatogr.* 19, 401-404 (2006). TLC of methionine, L-cystine, calcium pantothenate, vitamin B1, vitamin B7, and p-aminobenzoic acid on silica gel in a pre-saturated chamber with n-propanol - water - chloroform 5:2:1. Quantification of methionine was based on the oxidation and reaction with leuco xylene cyanol FF solution, which formed a blue dye. Quantitative determination by absorbance measurement at 613 nm.

quality control, qualitative identification, quantitative analysis 18a

## 21. Purines, pyrimidines, nucleic acids and their constituents

- 98 049 Jolanta SOCHACKA\*, A. KOWALSKA (\* Department of General and Analytical Chemistry, Faculty of Pharmacy, The Medical University of Silesia, Jagiellonska 4, 41-200 Sosnowiec, Poland): Comparison of calculated values of the lipophilicity of 2,6-disubstituted 7-methylpurines with values determined by RPTLC. *J. Planar Chromatogr.* 19, 307-312 (2006). TLC of 2,6-disubstituted 7-methylpurines and 6-mercaptopurine on RP-18 with mixtures of acetone and buffer (sodium acetate - veronal, pH 7.0). Detection in UV light at 254 nm. Partly good agreement was obtained between experimental log P (TLC) values and calculated Clog P values.

qualitative identification 21a

## 22. Alkaloids

- 98 050 N. C. NIKOLIC\*, M. Z. STANKOVIC (\*Fac. of Techn., Bulevar oslobodjenja 124, 16000 Leskovac, Serbia and Montenegro): Acid hydrolysis of potato tuber sprout glykoalkaloids and kinetics of solanidine extraction in three-phase systems. *Ital. J. Food Sci.* 18, 287-294 (2006). The ratio

of alpha-solanine to alpha-chaconine of tuber sprouts was determined by TLC on silica gel with methanol - chloroform - 1 % aqueous ammonium hydroxide 2:2:1. For quantitative determination zones were isolated from the plate, extracted with 2 % aqueous acetic acid, and measured spectrophotometrically at 420 nm with methyl orange.

food analysis, qualitative identification, autoradiography, potato 22

- 98 051 A. EVIDENTE\*, A. ANDOLFI, A. ABOU-DONIA, S. TOUEMA, H. HAMMODA, E. SHAW-KY, A. MOTTA (\*Dipartimento di Scienze del Suolo della Pianta e dell' Ambiente, Università di Napoli Federico II, Via Università 100, I-80055 Portici, Italy, evidente@unina.it): (-)-Amarbellisine, a lycorine-type alkaloid from *Amaryllis belladonna* L. growing in Egypt. *Phytochemistry*. 65, 2113-2118 (2004) . HPTLC of (-)-amarbellisine in the bulbs of *Amaryllis belladonna* L. on silica gel with chloroform - methanol 9:1 with 1 drop of ammonia. Quantitative determination at 254 nm to estimate the alkaloidal content in the flowering stage (in April), and in the preflowering stage (in November).

herbal, HPTLC, densitometry, quantitative analysis 22

- 98 052 M. GADZIKOWSKA\*, G. GRYNKIEWICZ (\*Dept. of Inorg. and Anal. Chem., Med. Acad., Staszica 6, 20-081 Lublin, Poland): Commentary on the chromatographic retention of *Chelidonium* alkaloids. *Acta Chrom.* 11, 62-74 (2001). TLC of basic isolates of *Chelidonium majus* L. (homochelidonine, allocryptopine, chelidonine, protopine, chelerythrine, sanguinarine and berberine) on silica gel with different mobile phases: toluene - ethyl acetate - methanol 17:2:1, toluene - ethyl acetate 4:1, toluene - ethyl acetate - methanol 11:8:1 and toluene - methanol 19:1; and on aluminum oxide 60 type E (basic) with toluene - ethyl acetate 9:1, 4:1, and 3:2; and toluene - methanol 19:1 in horizontal chamber. Evaluation under UV light at 254 and 366 nm, or by spraying with Dragendorff's reagent. The HPLC-UV method was developed in parallel.

herbal, qualitative identification, alkaloids, *Chelidonium*, qualitative analysis 22

- 98 053 M. M. GUPTA\*, A. SRIVASTAVA, A. K. TRIPATHI, H. MISRA, R. K. VERMA (\*Analytical Chemistry Division, Central Institute of Medicinal and Aromatic Plants, P. O. CIMAP, Lucknow-226015, India): Use of HPTLC, HPLC, and densitometry for qualitative separation of indole alkaloids from *Rauwolfia serpentina* roots. *J. Planar Chromatogr.* 19, 282-287 (2006). HPTLC of ajmaline, ajmalicine, and reserpine on silica gel, prewashed with methanol, in an unsaturated twin-trough chamber with toluene - methanol 19:6. Detection by dipping in Dragendorff's reagent. Quantitative determination by absorbance measurement at 520 nm.

herbal, traditional medicine, HPTLC, densitometry, quantitative analysis 22

- 98 054 Ágnes SÁRKÖZI\*, Á. M. MÓRICZ, P. G. OTT, E. TYIHÁK, Á. KÉRY (\*Department of Pharmacognosy, Faculty of Pharmacy, Semmelweis University, Üllői Str. 26, 1085 Budapest, Hungary): Investigation of *Chelidonium* alkaloids by use of a complex bioautographic system. *J. Planar Chromatogr.* 19, 267-272 (2006). TLC of chelerythrine, chelidonine, and sanguinarine on silica gel with dichloromethane - methanol 97:3. Use of the mobile phase recommended by Ph. Eur. 5, propanol - water - formic acid 90:9:1, did not enable satisfactory separation of *Chelidonium* alkaloids. Development at 20 - 24 °C and 60 % relative humidity in a presaturated TLC chamber. After drying evaluation under UV light at 254 and 366 nm.

herbal, quality control, qualitative identification, bioautography 22

### 23. Other substances containing heterocyclic nitrogen

- 98 055 Joanna NOWAKOWSKA (Medical University of Gdansk, Faculty of Pharmacy, Department of

Physical Chemistry, Al. Gen. J. Hallera 107, 80-416, Gdansk, Poland): The retention behavior of selected porphyrins on silica gel, polyamide, and cellulose TLC plates. *J. Planar Chromatogr.* 19, 393-397 (2006). TLC of uroporphyrin I, uroporphyrin III, coproporphyrin I, coproporphyrin III, and protoporphyrin IX (as methyl esters) on polyamide 11, cellulose and silica gel with methanol, ethanol, propanol, butanol, acetonitrile and tetrahydrofuran in a saturated chamber. Porphyrins on cellulose were detected by placing in iodine vapor for 5 min; on silica gel and polyamide 11 the porphyrins were detected as red spots in UV light at 254 nm. Chromatographic retention data and a possible retention mechanism are discussed.

clinical chemistry research, qualitative identification 23a

- 98 056 M. PODGORNA (Inst. of Chem., Silesian Univ., 9 Szkolna Street, 40-006 Katowice, Poland): Separation of porphyrins and metallporphyrins by TLC. *Acta Chrom.* 12, 226-233 (2002). TLC of porphyrins and their complexes with Cu (II) and Ni (II) on silica gel (activated at 120 °C for 30 min) with carbon tetrachloride - chloroform 1:1 after chamber saturation for 30 min. After drying the plates at RT, spots were visible on the plates. The effects of Cu (II) and Ni (II) cations on the separation of hydroxy and alkyloxy tetraphenylporphyrin derivatives were determined.

metallporphyrins 23a

## 27. Vitamins and various growth regulators

- 98 057 I. BARANOWSKA\*, A. KADZIOLKA (\*Dept. of Anal. and General Chem., Silesian Technical Univ., Gliwice, Poland): RPTLC and derivative spectrophotometry for the analysis of selected vitamins. *Acta Chrom.* 6, 61-71 (1996). TLC on RP-18 in Shandon chamber with water - methanol 5:4 and water - acetic acid 7:1 for separation of water-soluble vitamins (nicotinic acid, nicotinamide, C, B1 and rutin) and with acetonitrile - benzene - chloroform 10:10:1 for fat-soluble vitamins (A-acetate, E, E-acetate). Visualization of rutin with 25 % lead (II) acetate, of other water-soluble vitamins with potassium hexaiodoplatinate (IV) solution prepared by mixing 10 % potassium iodide with 5 % hexachloroplatinic acid 9:1. Fat-soluble vitamins were detected with a 10 % solution of antimony chloride. Derivative spectrophotometry was applied to the determination of vitamins B1, B6 and A-acetate in mixtures with other vitamins.

food analysis, qualitative identification, vitamins 27

## 28. Antibiotics, Mycotoxins

- 98 058 S. A. NABI\*, M. A. KHAN, S. N. KHOWAJA, A. ALIMUDDIN (\*Dept. of Chem., Aligarh Muslim Univ., Aligarh, India): Thin-layer chromatographic separation of penicillins on stannic arsenate-cellulose layers. *Acta Chrom.* 16, 164-172 (2006). TLC of ampicillin, amoxicillin, penicillamine, benzylpenicillin and penicillin in commercial drugs (megaphen and hipenox) on self-made stannic arsenate - cellulose 1:4 layers containing 10 % CaSO<sub>4</sub> as binder, with acetonitrile - acetic acid - chloroform 2:3:9, and acetone - acetic acid - chloroform 5:5:6. Detection by treatment with iodine vapour. The zones were extracted with methanol and spectrophotometrically determined at 460 nm using 2,3-dichloro-5,6-dicyano-p-benzoquinone as derivatization reagent.

pharmaceutical research, qualitative identification, penicillin 28a

## 29. Pesticides and other agrochemicals

- 98 059 S. A. NABI\*, A. GUPTA, M. A. KHAN, A. ISLAM (\*Anal. Chem. Div., Dept. of Chem., Aligarh Muslim Univ., Aligarh-202002, India): Thin-layer chromatographic separations of some common pesticides on mixed stannic oxide-silica gel G layers. *Acta Chrom.* 12, 201-210 (2002). TLC of important organophosphorus, organochlorine, and pyrethroid pesticides (dichlorvos, endosulfan, malathion, monocrotophos, parathion, phorate, phosphamidon, quinalphos, dimethoate, anilofos,



chlorpyrifos, fenvalerate, methyl oxydemeton, and methyl parathion) on stannic oxide-silica gel G layers with a variety of mixed aqueous and organic mobile phases. Detection by treatment with iodine vapor. For quantification corresponding zones were removed from the plate and extracted with ethanol. Quantitative determination by spectrophotometry with the molybdenum blue method. Monocrotophos, dimethoate, and malathion in soil samples were analyzed to test the applicability of the simple method.

environmental, qualitative identification, pesticides

29b

- 98 060 S. A. NABI\*, A. SIKARWAR (\*Anal. Research Lab., Dept. of Chem., Aligarh Muslim Univ., Aligarh-202002, India) : Separation of phenolic compounds on stannic arsenate - silica gel layers - quantitative separation and determination of phenolic pesticide residues in bananas. *Acta Chrom.* 9, 123-132 (1999). TLC of phenolic pesticide residues (m-nitrophenol, p-nitrophenol and p-aminophenol) in banana fruits and plant-tissues on stannic arsenate-silica gel layers. Several mobile phases of different polarity were evaluated; separation of m-nitrophenol, p-nitrophenol, picric acid and p-aminophenol was best achieved with ethanol - 1.0 M citric acid 1:3. Visualization by spraying with AgNO<sub>3</sub> - reagent (saturated AgNO<sub>3</sub> solution in acetone 1:20 diluted with water until the precipitate of AgNO<sub>3</sub> was dissolved) followed by heating at 105 °C, or by spraying with a mixture of 15 % FeCl<sub>3</sub> and 1 % K<sub>4</sub>Fe(CN)<sub>6</sub> 1:1.

qualitative identification, pesticide residues

29, 32e

- 98 061 B. B. DAUNDKAR, R. R. MAVLE, M. K. MALVE, R. KRISHNAMURTHY\* (\*Directorate of Forensic Science Laboratories, State of Maharashtra, Hans Bhugra Marg, Kalina, Vidyanagari, Santa Cruz (E), Mumbai-400098, India): Detection of carbaryl insecticide in biological samples by TLC with a specific chromogenic reagent. *J. Planar Chromatogr.* 19, 467-468 (2006). TLC of carbaryl (1-naphthylmethylcarbamate), a common derivative of the widely used N-methylcarbamate insecticide, on silica gel with n-hexane - acetone 4:1 in a saturated twin-trough chamber. After drying the plate was heating at 100 °C for 5 min, cooled, and sprayed successively with 5 % sodium hydroxide solution and with a 1:1 mixture of 2 % diphenylamine solution and 5 % formaldehyde solution. Visual detection of the blue-green zones carbaryl.

toxicology, qualitative identification

29c

- 98 062 J. ESPINOZA, M. BAEZ\* (\*Departamento de Química Inorgánica y Analítica, Facultad de Ciencias Químicas y Farmacéuticas, Universidad de Chile, Casilla 233, Santiago, Chile): Determination of atrazine in aqueous soil extracts by high performance thin-layer chromatography. *J. Chil. Chem. Soc.* 48, 19-23 (2003). HPTLC validation of atrazine in aqueous extracts of soils on silica gel (layer thickness 100 µm) previously activated at 120 °C for 30 min. The elution program applied to aqueous soil matrices started with 10 short isocratic runs (0.8 min) with acetonitrile - dichloromethane 30:70. Mixer was emptied after the tenth step and refilled to continue with 4 successive isocratic runs (2.5, 5.0, 7.5 and 25 min) with dichloromethane. The plate was dried for 1 min between each step and for 3 min after the last one. The plate was preconditioned with nitrogen for 15 s before each run. Quantitative determination by absorbance at 210 nm. Linearity is between 5 and 100 ng and recoveries ranging from 98.7 to 103.5 %. The detection limit is 1.5 ng and the quantification limit is 4.9 ng. Precision analysis shows an intra-assay variation between 1.48 and 5.47 %. The method can be applied to a broad range of soils including those with high organic matter content.

environmental, HPTLC, densitometry, quantitative analysis, AMD

29d

- 98 063 Malgorzata JANICKA (Faculty of Chemistry, Department of Physical Chemistry, Maria Curie-Skłodowska University, M. Curie-Skłodowska Sq. 3, 20-031 Lublin, Poland): Comparison of

different properties - log P, log kW, and  $\phi_0$  - as descriptors of the hydrophobicity of some fungicides. *J. Planar Chromatogr.* 19, 361-370 (2006). Log P, log kW, and  $\phi_0$  are proposed and compared as descriptors of the hydrophobicity of twenty-two dihydroxythiobenzanilides. Chromatographic data log kW and  $\phi_0$ , were calculated from experimental TLC results obtained on RP-18W and cyano phases with methanol or acetone as organic modifiers. The calculated hydrophobicity was compared with the biological activity of the test substances. Development in saturated sandwich chambers at 20 °C. Detection at 320 nm.

agricultural, qualitative identification

29e

### 30. Synthetic and natural dyes

98 064 J. SECHRIST, J. PACHUSKI, J. SHERMA\* (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA): Quantification of lutein in dietary supplements by reversed-phase high-performance thin-layer chromatography with visible-mode densitometry. *Acta Chrom.* 12, 151-158 (2002). HPTLC of lutein in nutrition supplements on RP-18 (with concentration zone, pre-washed with dichloromethane - methanol 1:1) with petroleum ether - acetonitrile - methanol 1:2:2. All sample solutions and developing chambers were wrapped in aluminum foil to protect the labile pigment from photo-decomposition. Densitometry at 448 nm. Three products containing lutein as the free alcohol or dipalmitate ester plus other ingredients were analyzed.

herbal, food analysis, quantitative analysis, HPTLC, densitometry, lutein

30b

### 32. Pharmaceutical and biomedical applications

98 066 G. AKOWUAH\*, I. ZHARI, I. NORHAYATI, A. MARIAM (\*Herbal secretariat, School of Pharmaceutical Sciences, Universiti Sains Malaysia, 11800 Minden, Penang, Malaysia, wuahmy@yahoo.com): HPLC and HPTLC densitometric determination of andrographolides and antioxidant potential of *Andrographis paniculata*. *J. Food Comp. Anal.* 19, 118 - 126 (2006). HPTLC of andrographolide (AP) and 14-deoxy-11,12-didehydroandrographolide (DIAP) in the aerial parts of *Andrographis paniculata* Nees on silica gel with chloroform - methanol 4:1 with chamber saturation for 2 h. Quantitative determination at 254 nm. Good resolution of AP and DIAP was obtained together with symmetrical and reproducible peaks at Rf 0.55 and 0.43, respectively. Linearity is between 10 and 2000 µg/mL; LOD is 3.0 and 3.6 µg/mL; mean recoveries are 97.7 % and 97.8 % and precision analysis shows an intra-assay variation between 0.89 - 0.99 % and an inter-assay variation between 0.86 - 0.98 %. HPTLC method leads to accurate results when compared to the HPLC method.

traditional medicine, HPTLC, quantitative analysis, densitometry

32e

98 067 S. ANANDJIWALA, J. KALOLA, M. RAJANI\* (\*B. V. Patel Pharmaceutical Education and Research Development Centre, Pharmacognosy and Phytochemistry Department, Thaltej, Ahmedabad-380 054, Gujarat, India; rajanivenkat@hotmail.com): Quantification of eugenol, luteolin, ursolic acid, and oleanolic acid in black (Krishna Tulasi) and green (Sri Tulasi) varieties of *Ocimum sanctum* Linn. using high-performance thin-layer chromatography. *J. Assoc. Off. Anal. Chem.* 89, 1467-1474 (2006). HPTLC of eugenol, luteolin, ursolic acid, and oleanolic acid on silica gel in a twin trough chamber with toluene - ethyl acetate - formic acid 35:15:1 at 25 °C and 40 % relative humidity. Quantitation in absorbance mode at 280 nm for eugenol, at 350 nm for luteolin, and at 530 nm for ursolic acid and oleanolic acid after derivatization with anisaldehyde-sulfuric acid reagent and heating at 105°C. The methods were validated for precision, repeatability, and accuracy.

traditional medicine, herbal, HPTLC, densitometry

32e

- 98 068 S. BABU, K. KUMAR, G. SUBBARAJU\* (\*Laila Impex Research Center, Vijayawada, India, subbarajugottumukkala@hotmail.com) : Estimation of trans-resveratrol in herbal extracts and dosage forms by high-performance thin-layer chromatography. *Chem. Pharm. Bull.* 53, 691-693 (2005). HPTLC of trans-resveratrol in the roots of *Polygonum cuspidatum* and in dosage forms on silica gel with chloroform - ethyl acetate - formic acid 25:10:1. Quantitative determination by absorbance measurement at 313 nm. Linearity of determination of trans-resveratrol is between 0.5 and 3.0 µg with a correlation coefficient of 0.9989. LOD is 9 ng, and LOQ is 27 ng. The average percentage recovery is 99.9 - 100.7 %.
- pharmaceutical research, quality control herbal, HPTLC, quantitative analysis, densitometry 32a
- 98 069 S. BADAMI\*, M. GUPTA, S. RAMASWAMY, S. RAY, M. NANJAIN, D. BENDELL, R. SUBBAN, S. BHOJARAJ (\*J.S.S. College of Pharmacy, Ootacamund, India, shribadami@yahoo.com): Determination of betulin in *Grewia tiliaefolia* by HPTLC. *J. Sep. Sci.* 27, 129-131 (2004). HPTLC of betulin in the bark of *Grewia tiliaefolia* Vahl. on silica gel with toluene - ethyl acetate 9:1. Quantitative determination by absorbance measurement. Linearity of determination of betulin is between 1000 and 1800 ng and its average percentage recovery is 96.09 - 98.87 %.
- herbal, quality control, densitometry, quantitative analysis, HPTLC 32g
- 98 070 A. BAFNA, S. MISHRA\* (\*Department of Pharmacy, Faculty of Technology and Engineering, The M.S. University of Baroda, Gujarat, India, shmishra48@rediffmail.com) : Protective effect of bioactive fraction of *Sphaeranthus indicus* Linn. against cyclophosphamide induced suppression of humoral immunity in mice. *J. Ethnopharmacol.* 104, 426-429 (2006). HPTLC of the methanol fraction (bioactive fraction) of the fresh flower heads of *Sphaeranthus indicus*, on silica gel with three different solvent systems: 1) toluene - ethyl acetate 7:3; 2) ethyl acetate - methanol - water 200:27:20; and 3) n-butanol - glacial acetic acid 3:1:1. Detection by spraying with flavone reagent and qualitative determination by absorbance measurement at 254 nm and by fluorescence measurement at 366 nm. The compound having Rf 0.07 reacted to the reagent in solvent system 1. Two compounds at Rf 0.09 and 0.36 reacted in solvent system 2, whereas five compounds having Rf 0.11, 0.35, 0.38, 0.49 and 0.72 reacted to the reagent in solvent system 3.
- herbal, traditional medicine, qualitative identification, HPTLC 32e
- 98 071 G. BALOGH, E. CSIZÉR, G. G. FERENCZY, Z. HALMOS, B. HERÉNYI, P. HORVATH, A. LAUKÓ, S. GÖRÖG\* (\*Chemical Works of Gedeon Richter Ltd. , P. O. B. 27, H-1475, Budapest, Hungary): Estimation of impurity profiles of drugs and related materials. 12. Isolation and identification of an isomeric impurity in danazol. *Pharm. Research* 12, 295-298 (1995). TLC of isodanazol according to The United States Pharmacopoeia XXII, 1990, pp. 379-380 and K. Ferenczi-Fodor, Z. Végh, Z. Pap-Sziklay, *J. Planar chromatogr.* 6, 198-203 (1993). Ferenczi-Fodor described a validated method for the selective determination of three individual impurities in danazol based on TLC-densitometric measurement at 252 nm. The Rf values of danazol and isodanazol in the TLC systems of USP XXII are 0.29 and 0.33, respectively, while in the system of Ferenczi-Fodor et al. 0.47 and 0.58.
- quality control, densitometry 32a
- 98 072 R. BHUSHAN\*, D. GUPTA, A. JAIN (\*Department of Chemistry, Indian Institute of Technology Roorkee, Roorkee-247 667, India): TLC supplemented by UV spectrophotometry compared with HPLC for separation and determination of some antidiabetic drugs in pharmaceutical preparations. *J. Planar Chromatogr.* 19, 288-296 (2006). TLC of metformin hydrochloride, pioglitazone hydrochloride, rosiglitazone maleate, gliclazide, and glibenclamide on silica gel in a pre-equilib-

rated chamber at 26 - 30 °C with toluene - ethyl acetate - methanol 17:2:17 and n-butanol - acetic acid - water - methanol 12:4:1:2. Detection by exposure to iodine vapor. Highly reproducible RF and compact spots were obtained in normal-phase TLC which was found to be the least expensive method.

quality control, comparison of methods

32a

- 98 095 A. N. CAMPBELL, J. SHERMA\* (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA) : Development and validation of a high-performance thin-layer chromatographic method with densitometric detection for determination of bisacodyl in pharmaceutical tablets. *Acta Chrom.* 13, 109-116 (2003). HPTLC of the laxative bisacodyl in enteric-coated tablets on silica gel with concentrating zone and 19 channels, pre-cleaned with dichloromethane - methanol 1:1, with ethyl acetate-methanol-glacial acetic acid 17:2:1. Quantitative determination by absorbance measurement at 254 nm. The method was validated and can be used for routine analysis of the pharmaceutical preparation in industry quality control and regulatory laboratories. An alternative extraction procedure and mobile phase are suggested for analysis of bisacodyl tablets with different formulations.

pharmaceutical research, quality control, HPTLC, qualitative identification, quantitative analysis, bisacodyl

32a

- 98 111 V.V. DIGHE, R. T. SANE, S.N. MENON, H. N. TAMBE\*, S. PILLAI (\*TDM Laboratories, Plot No. 194, Scheme No. 6, Road No. 15, Sion (E), Koliwada, Mumbai-22, India): High-performance thin-layer chromatographic determination of itopride hydrochloride in its pharmaceutical preparation and in the bulk drug. *J. Planar Chromatogr.* 19, 319-323 (2006). HPTLC of itopride hydrochloride on silica gel with methanol - ethyl acetate - toluene - triethylamine 2:5:12:1. Quantitative determination by absorbance measurement at 230 nm. The method was validated for accuracy and precision.

quality control, HPTLC, quantitative analysis, densitometry

32a

- 98 074 V.V. DIGHE, R. T. SANE, S. N. MENON, H. N. TAMBE\*, S. PILLAI, V. N. GOKARN (\*TDM Laboratories, Plot No. 194, Scheme No. 6, Road No. 15, Sion (E), Sion Koliwada, Mumbai-22, India): Simultaneous determination of diclofenac sodium and paracetamol in a pharmaceutical preparation and in bulk drug powder by high-performance thin-layer chromatography. *J. Planar Chromatogr.* 19, 443-448 (2006). HPTLC of diclofenac sodium and paracetamol (with aceclofenac as internal standard) on silica gel, pre-washed with methanol, in a presaturated twin-trough chamber with toluene - ethyl acetate - methanol - formic acid 50:40:10:1. Quantitative determination by absorbance measurement at 260 nm. The method was validated regarding accuracy and precision.

quality control, HPTLC, densitometry, quantitative analysis

32a

- 98 087 D. M. DIGREGORIO\*, H. D. HARNETT, J. SHERMA (\*Dept. of Chem., Lafayette College, Easton, PA 18042, USA): Quantification of dextromethorphan hydrobromide and clemastine fumarate in pharmaceutical caplets, gelcaps and tablets by HPTLC with ultraviolet absorption densitometry. *Acta Chrom.* 9, 72-78 (1999). The method is developed for determination of the cough suppressant dextromethorphan hydrobromide in multi-component flu-relief solid caplets and liquid gel caps, and of the antihistamine clemastine fumarate in tablets. HPTLC on silica gel (pre-washed with dichloromethane - methanol 1:1) with ethyl acetate - methanol - ammonia 17:1:2 for dextromethorphan hydrobromide or dichloromethane - methanol - ammonia 90:10:1 for clemastine fumarate. Densitometry at 225 nm for dextromethorphan hydrobromide and at 216 nm for clemastine fumarate. Multiple samples of the three dosage forms were analyzed to

confirm agreement between content of the active ingredients and label declarations.

pharmaceutical research, quantitative analysis, densitometry 32a

- 98 075 M. FAN (Fan Minwei), (Shanghai Municipal Inst. TCM, Shanghai 200127, China): (Study of the quality standard of Ruyi Jinhuang Tie ointment) (Chinese). *J. Chinese Trad. Patent Med. (Zhongchengyao)* 27(7), 864-867 (2005). TLC of extracts of Ruyi Jinhuang Tie ointment on silica gel with 1) n-butanol - glacial acetic acid - water 7:1:2; 2) petroleum ether (30 - 60 °C) - ethyl formate - formic acid 15:5:1; 3) toluene - ethyl acetate - acetic acid 13:9:1; 4) chloroform - methanol - formic acid 96:3:1; 5) ethyl acetate - methanol - water 100:17:3 and toluene - ethyl acetate - formic acid water 20:10:1; 6) n-hexane - benzene - ethyl acetate 14:3:3. Detection under UV 365 nm and by spraying with 1 % FeCl<sub>3</sub> solution in ethanol or with a solution of 5 % p-dimethylaminobenzaldehyde 10 % ethanolic H<sub>2</sub>SO<sub>4</sub> followed by heating at 80 °C. Identification of the fingerprint. Quantitative determination of berberine chloride by HPLC. The results of three batches of real life samples are given.

pharmaceutical research, traditional medicine, quality control, qualitative identification, berberine chloride 32c

- 98 076 A. FAYED\*, M. SHEHATA, N. HASSAN, S. EL-WESHAHY (\*Analytical Chemistry Department, Faculty of Pharmacy, Cairo University, Cairo, Egypt, fayedaeg@yahoo.com): Validated HPLC and HPTLC stability-indicated methods for determination of alfuzosin hydrochloride in bulk powder and pharmaceutical formulations. *J. Sep. Sci.* 29, 2716-2724 (2006). HPTLC of alfuzosin hydrochloride subjected to stress conditions (alkaline, acidic, oxidative, thermal, and photo-degradation) on silica gel with methanol - ammonia 125:3. Quantitative determination by absorbance measurement at 245 nm. Linearity of determination of alfuzosin is between 0.5 and 7.0 µg and its average percentage recovery is 98.5 - 101.8 %. The drug is well separated from its degradation products upon applying the two methods.

pharmaceutical research, HPTLC, densitometry, quantitative analysis, comparison of methods 32a

- 98 077 J. FISHER, J. SHERMA\* (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA): Analysis of carisoprodol tablets by HPTLC with visible absorbance densitometry. *Acta Chrom.* 11, 96-101 (2001). HPTLC of the muscle relaxant carisoprodol on silica gel uniplates with inorganic binder and fluorescent indicator, prewashed with dichloromethane - methanol 1:1, with chloroform - acetone 4:1 as mobile phase. Detection by spraying with conc. sulfuric acid - methanol 1:1 followed by heating at 150 °C for 5 min. Quantitative determination by absorbance measurement at 550 nm. The method was applied to tablets containing carisoprodol as the only active ingredient and to tablets containing carisoprodol with aspirin and with aspirin plus codeine phosphate.

densitometry, quantitative analysis, HPTLC, carisoprodol 32a

- 98 078 Jolanta FLIEGER\*, M. TATARCZAK (\*Department of Inorganic and Analytical Chemistry, Medical University of Lublin, Staszica 6, 20-081 Lublin, Poland): Effect of inorganic salts as mobile-phase additives on lipophilicity values determined by reversed-phase thin-layer chromatography for new 1,2,4-triazole derivatives. *J. Planar Chromatogr.* 19, 386-392 (2006). TLC of 11 new 1,2,4-triazole derivatives on RP-18 with mixtures of methanol and water containing from 20 to 70 % methanol in steps of 10 %. Mobile phases containing different additives were prepared by adding the sodium salts of different anions (sodium hexafluorophosphate, perchlorate, trifluoroacetate, nitrate, chloride, iodide, and dihydrophosphate) to the mobile phase. Addition of iodide anions proved a key factor in obtaining lipophilicity indexes which correlated better with the log PG scale for all solutes investigated.

- pharmaceutical research, qualitative identification 32a
- 98 065 S. A. GOSAVI, A. A. SHIRKHEDKAR\*, Y. S. JAISWAL, S. J. SURANA (\*Department of Pharmaceutical Chemistry, R. C. Patel College of Pharmacy, Karwand Naka, Shirpur-Dhule, M. S.- (425405), India): Quantitative planar chromatographic analysis of pantoprazole sodium sesquihydrate and domperidone in tablets. *J. Planar Chromatogr.* 19, 302-306 (2006). HPTLC of pantoprazole sodium sesquihydrate and domperidone on silica gel, prewashed with methanol, at 23 - 27 °C in a pre-saturated twin-trough chamber with methanol - water - ammonium acetate 8:2:1. Quantitative determination by absorbance measurement at 286 nm. The method was validated in accordance with ICH guidelines.  
quality control, HPTLC, densitometry 32a
- 98 079 Z. HASSANKHANI-MAJD, V. GHOULIPOUR, S. W. HUSAIN\* (\*Anal. Lab., Dept. of Applied Chem., Fac. of Chem., Univ. of Tarbiat Moallem, 49 Mofatteh Avenue, Tehran-15614, Iran): Chromatographic behaviour of performance-enhancing drugs on thin layers of bismuth silicate ion exchanger. *Acta Chrom.* 16, 173-180 (2006). TLC of 14 performance-enhancing drugs (amphetamine, bemegride, caffeine, chlorphentermine, ephedrine, ethylamphetamine, isoproterenol, methadone, methylendioxyamphetamine, pentazocine, pethidine, pemoline, strychnine and salbutamol) on bismuth silicate gel (prepared from 75 mL bismuth nitrate gel with 14 g silica gel powder) with thickness of 300 µm. 21 organic, aqueous and organic-aqueous mobile phases were investigated. Detection with iodine vapours.  
doping, qualitative identification 32c
- 98 080 X. HOU (Hou Xiuzhen)\*, X. QI (Qi Xihong) (\*Ningxia Inst. Drug Cont., Yinchuan, Ningxia 750004, China): (Determination of betaine in *Lycium barbarum* L. by thin-layer chromatography) (Chinese). *J. Chinese Trad. Patent Med. (Zhongchengyao)* 27 (8), 941-944 (2005). TLC of betaine on silica gel with acetone - ethanol - hydrochloric acid 10:6:1. Detection by spraying with a solution of potassium iodobismuthate. Quantitative determination by densitometry at 515 nm. Validation regarding linearity range (8.0 - 39.0 µg,  $r = 0.9992$ ), precision (RSD = 1.62 %,  $n = 6$ ), reproducibility (RSD = 2.14 %,  $n = 6$ ), and recovery (99.75 %, RSD = 1.92 %,  $n = 6$ ). Results are given for 12 real life samples. Discussion of the advantages of the method compared to HPLC.  
pharmaceutical research, quality control, HPTLC, densitometry, quantitative analysis, qualitative identification, betaine 32c
- 98 081 X. JIA (Jia Xiansheng)\*, H. WU (Wu Hongfei), Q. MAO (Mao Qing) (\*Guiyang Coll. TCM, Guiyang 550002, China): (Determination of fulvotomentoside A in *Lonicera Fulvotomentosa* Hsu et S. C. Cheng by thin-layer chromatography) (Chinese). *Chinese J. Pharm. Anal. (Yaowu Fenxi Zazhi)* 25 (6), 719- 721 (2005). TLC of fulvotomentoside A on silica gel with chloroform - methanol - water 61:32:5. Detection by spraying with 10 % H<sub>2</sub>SO<sub>4</sub> in ethanol followed by heating at 105 °C for 5 min. Quantitative determination by densitometry at 518 nm. The procedure was validated regarding linearity range (1.20 - 7.20 µg/spot,  $r = 0.998$ ), repeatability (0.51 %,  $n = 5$ ), precision (0.21 %,  $n = 5$  within plate and 0.87 %  $n = 5$  plate-to-plate), and recovery (99.7 %, RSD = 2.45 %,  $n = 5$ ). The results are given for three batches of real life samples.  
pharmaceutical research, herbal quality control, traditional medicine, qualitative identification, quantitative analysis, densitometry, fulvotomentoside A 32c
- 98 084 P. KANETKAR, R. SINGHAL, K. LADDHA, M. KAMAT\* (\*Food Engineering and Technology Department, Institute of Chemical Technology, University of Mumbai, Mumbai, India, mykamat@udct.org): Extraction and quantification of gymnemic acids through gymnemagenin

from callus cultures of *Gymnema sylvestre*. *Phytochem. Analysis* 17, 409-413 (2006). HPTLC of gymnemic acid as of gymnemagenin from callus cultures of *Gymnema sylvestre* on silica gel with chloroform - methanol 4:1. Quantitative determination by absorbance measurement at 205 nm. Linearity of determination of gymnemagenin is between 2 and 10 µg and its average percentage recovery from leave callus extracts is 98.6 - 99.2 %.

herbal, quantitative analysis, HPTLC

32e

- 98 085 N. KAUL, H. AGRAWAL, P. MASKE, J. RAO\*, K. MAHADIK, S. KADAM (\*Department of Quality Assurance Techniques, Bharati Vidyapeeth Deemed University, Maharashtra State, India, janhavirao@rediffmail.com) : Chromatographic determination of itopride hydrochloride in the presence of its degradation products. *J. Sep. Sci.* 28, 1566-1576 (2005). HPTLC of itopride (subjected to acid and alkaline hydrolysis, oxidation, dry and wet heat treatment, UV and photo-degradation) on silica gel with toluene - methanol - chloroform 5:3:6 and 1 drop of ammonia. Quantitative determination by absorbance measurement at 270 nm. The drug is well separated from its degradation products upon applying the two methods.

pharmaceutical research, HPTLC, densitometry, quantitative analysis, comparison of methods

32a

- 98 073 Y. C. KUO (Yuh-Chi Kuo), C. K. LU (Chung-Kung Lu), L. W. HUANG (Li-Wei Huang), Y. H. KUO (Yueh-Hsiung Kuo), C. CHANG (Chen Chang), F. L. HSU (Feng-Liu Hsu), T. H. LEE\* (Tzong-Huei Lee) (\*Graduate Institute of Pharmacognosy, Taipei Medical University, Taipei, Taiwan 110, Republic of China; e-mail: thlee@tmu.edu.tw): Inhibitory effects of acylated kaempferol glycosides from the leaves of *Linnamomum kotoense* on the proliferation of human peripheral blood mononuclear cells. *Planta Med.* 71, 412-415 (2005). TLC of kaempferol 3-O-alpha-L-[2-(Z)-p-coumaroyl-4-(E)-p-coumaroyl]rhamnopyranoside and kaempferol 3-O-alpha-L-[2,4-di-(E)-p-coumaroyl]rhamnopyranoside on silica gel with dichloromethane - methanol 6:1. Observation under UV light at 254 nm and dipping in vanillin - sulfuric acid resulting in yellow-green spots.

traditional medicine, herbal, qualitative identification

32e

- 98 086 L. LIU (Liu Lifang)\*, SH. CAO (Cao Shuping), Y. JI (Ji Ying), X. ZHANG (Zhang Xiaojun), T. HUANG (Huang Ting) (\*China Pharm. Univ., Nanjing 210038, China): Qualitative and quantitative analysis of aristolochic acids in Chinese materia medica and traditional Chinese patent medicines. *J. Chinese Trad. Patent Med. (Zhongchengyao)* 27 (8), 938-941 (2005). TLC on silica gel with ethyl acetate - water - formic acid 10:1:1. Detection under UV 254 nm. Quantification of aristolochic acid I and II by HPLC with gradient elution. Results are given for six batches of real life samples.

pharmaceutical research, traditional medicine, quality control, qualitative identification, aristolochic acids

32c

- 98 088 X. MAO (Mao Xiuhong)\*, K. WANG, SH. JI ( Ji Shen) (\*Shanghai Municipal Inst. Drug Cont., Shanghai 200233, China): (Study of quality standard for Wumei Rendan pills) (Chinese). *J. Chinese Trad. Patent Med. (Zhongchengyao)* 27 (8), 896-899 (2005). TLC of Wumei Rendan pills on silica gel with ethyl acetate - formic acid - water 2:1:1. Detection by spraying with 10 % ethanolic H<sub>2</sub>SO<sub>4</sub> solution followed by heating at 105 °C. Identification of the fingerprint. Quantification of menthol by GC, and of ammonium glycyrrhizate by HPLC.

pharmaceutical research, traditional medicine, quality control, qualitative identification, menthol, ammonium glycyrrhizate

32c

- 98 089 W. MARKOWSKI\*, K. L. CZAPINSKA, G. MISZTAL, L. KOMSTA (\*Department of Physical Chemistry, Medical University Lublin, Staszica St. 6, 20-081 Lublin, Poland): Analysis of some fibrate-type antihyperlipidemic drugs by AMD. *J. Planar Chromatogr.* 19, 260-266 (2006). AMD-HPTLC of bezafibrate, ciprofibrate, clofibric acid, clofibrate, fenofibrate, and gemfibrozil on diol phase with mixtures of tetrahydrofuran and hexane. An 25 % aqueous solution of acetic acid was used for preconditioning. Quantitative determination by absorbance measurement at 227 and 254 nm. quality control, AMD, densitometry, HPTLC 32a
- 98 090 S. MENNICKENT\*, M. VEGA, C. GODOY (\*Departamento de Farmacia, Facultad de Farmacia, Universidad de Concepción, Casilla 237, Concepción, Chile, smennick@udec.cl): Development and validation of a method using instrumental planar chromatography for quantitative analysis of carbamazepine in saliva. *J. Chil. Chem. Soc.* 48, 71-73 (2003). HPTLC validation of carbamazepine in saliva samples on silica gel previously activated at 130 °C for 20 min. Development over 5 cm in a saturated chamber with ethyl acetate - toluene - methanol 5:4:1. Detection by dipping in 60 % perchloric acid in ethanol - water 1:1, followed by heating at 120°C for 7 min. Quantitative determination by fluorescence measurement at 366 nm. Linearity is between 0.5 and 15.0 ng per spot. The detection limit is 0.18 ng and the quantification limit is 0.54 ng. Precision: The analysis shows an intra-assay variation between 5.1 - 7.4 % and an inter-assay variation between 5.6 - 7.4 %. The method allows separation of carbamazepine from its main metabolites 10,11-dihydrocarbamazepine and carbamazepine-10,11-epoxide. clinical routine analysis, HPTLC, densitometry, quantitative analysis 32c
- 98 091 S. MEYYANATHAN\*, G. RAMASARMA, B. SURESH (\*Department of Pharmaceutical Chemistry, J.S.S. College of Pharmacy, Ootacamund, Tamilnadu, India, meyyys@rediffmail.com): Analysis of levofloxacin in pharmaceutical preparations by high performance thin layer chromatography. *J. Sep. Sci.* 26, 1698-1700 (2003). HPTLC of levofloxacin and lamotrigine (internal standard) on silica gel with water - methanol - n-butanol 1:1:1 and 1 drop of ammonia. Quantitative determination by absorbance measurement at 298 nm. Linearity of determination of levofloxacin is between 0.8 and 3.0 µg and its average percentage recovery is 99.9 %. pharmaceutical research, quality control, HPTLC, densitometry, quantitative analysis 32a
- 98 092 S. MEYYANATHAN\*, P. KUMAR, B. SURESH (\*J.S.S. College of Pharmacy, Ootacamund, Tamilnadu, India, meyyys@rediffmail.com): Analysis of tramadol in pharmaceutical preparations by high performance thin layer chromatography. *J. Sep. Sci.* 26, 1359-1362 (2003). HPTLC of tramadol and chlorzoxazone (internal standard) on silica gel with ethyl acetate - methanol 7:1 and 1 drop of ammonia. Quantitative determination by absorbance measurement at 275 nm. Linearity of determination of levofloxacin is between 1.0 and 2.5 µg and its average percentage recovery is 104.6 %. pharmaceutical research, quality control, HPTLC, quantitative analysis 32a
- 98 093 Zlata MRKVICKOVÁ\*, P. KOVANKOVÁ, J. KLIMES, M. DOLEZAL (\*Faculty of Pharmacy in Hradec Králové, Charles University in Prague, Heyrovského 1203, 500 05 Hradec Králové, Czech Republic): Determination of the lipophilicity of potential antituberculous compounds by RP-TLC. *J. Planar Chromatogr.* 19, 422-426 (2006). TLC of 26 substituted anilides of pyrazine-2-carboxylic acids on silica gel impregnated with a solution of silicone oil in diethyl ether 19:1 for 16 h. The aqueous component of the mobile phases was 0.05 M phosphate buffer of pH 7.4 or 3.0; methanol was used as organic modifier; separation in a pre-saturated chamber. Detection under UV light at 254 nm. . pharmaceutical research, qualitative identification 32a



- 98 094 Dragana MUTAVDZIC\*, S.BABIC, D. ASPERGER, A. J. M. HORVAT, M. KASTELAN-MACAN (\*Faculty of Chemical Engineering and Technology, Laboratory of Analytical Chemistry, Marulicev trg 19, 10000 Zagreb, Croatia): Comparison of different solid-phase extraction materials for sample preparation in the analysis of veterinary drugs in water samples. *J. Planar Chromatogr.* 19, 454-462 (2006). HPTLC of enrofloxazine, norfloxazine, oxytetracycline, trimethoprim, sulfamethazine, sulfadiazine, and penicillin G/procaine on cyano phase with 0.05 M oxalic acid - methanol 81:19. Evaluation under UV light at 254 and 366 nm. Quantitative determination of TMP by absorbance measurement at 254 nm, and fluorescence measurement at 366 nm for the other compounds. The SPE-TLC determination was validated for linearity, precision, quantification, and detection limit.  
agricultural, HPTLC, densitometry 32a
- 98 096 N. NADER\*, S. ESMAEILI, F. NAGHIBI, M. MOSADDEGH (\*Traditional Medicine and Materia Medica Research Center, Shaheed Beheshti University of Medical Sciences, P. O. Box 14155-6354, Tehran, Iran): HPTLC determination of apigenin in some Iranian liquid products of *Matricaria chamomilla* L.. *J. Planar Chromatogr.* 19, 383-385 (2006). HPTLC of apigenin on silica gel, pre-washed with methanol, in a saturated twin-trough chamber with toluene - methanol 5:1. Densitometric evaluation at 343 nm.  
herbal, traditional medicine, HPTLC densitometry, qualitative identification 32e
- 98 097 O. POZHARITSKAYA, S. IVANOVA, A. SHIKOV\*, V. MAKAROV (\*Interregional Center "Adaptogen", Std Petersburg, Russia, alexs79@mail.ru): Separation and quantification of terpenoids of *Boswellia serrata* Roxb. extract by planar chromatography techniques (TLC and AMD). *J. Sep. Sci.* 29, 2245-2250 (2006). HPTLC of four boswellic acids: 11-keto-beta-boswellic acid (1), acetyl-11-keto-beta boswellic acid (2), beta-boswellic acid (3), and acetyl-beta-boswellic acid (4) on silica gel with automated multiple development (AMD) using solvent gradients. Quantitative determination of 1 and 2 by absorbance measurement at 254 nm. 3 and 4 are quantified after derivatization with anisaldehyde sulfuric acid reagent at 560 nm. The AMD system provides good separation and the method is simple, precise, specific, sensitive, and accurate.  
herbal, HPTLC, densitometry quantitative analysis, AMD, postchromatographic derivatization 32g
- 98 098 V. PURATCHIMANI, S. JHA\* (\*Department of Pharmaceutical Sciences, Birla Institute of Technology, Mesra, Ranchi-835215, India; puratchi\_v@rediffmail.com): Standardisation of *Gymnema sylvestre* R. Br. with reference to gymnemagenin by high-performance thin-layer chromatography. *Phytochem. Anal.* 17, 164-166 (2006). HPTLC of gymnemagenin on silica gel with chloroform - methanol 9:1. Quantitative determination by absorbance measurement at 290 nm. Linearity of the determination of gymnemagenin was observed in the range of 4 - 10 µg. The average percentage recovery from an extract was 99.1 %, the content of leaves was 1.61 % (dry weight).  
quality control, herbal, HPTLC, densitometry 32e
- 98 099 Alina PYKA\*, M. BABUSKA, J. SLIWIOK (\*Department of Analytical Chemistry, Faculty of Pharmacy, Silesian Academy of Medicine, 4 Jagiellonska Street, 41-200 Sosnowiec, Poland): Use of liquid chromatography and theoretical computational methods to compare the lipophilicity of selected cortison derivatives. *J. Planar Chromatogr.* 19, 432-437 (2006). TLC of cortisone derivatives (corticosterone acetate, 11-dehydrocorticosterone acetate, corticosterone, 11-dhydrocorticosterone, allo-dihydrocortisone, hydrocortisone, and cortisone) on a mixture of silica gel and kieselguhr impregnated with a 10 % solution of paraffin oil in hexane. The plates were developed

with methanol - water 3:2. Quantitative determination by absorbance measurement at 254 nm.  
pharmaceutical research, qualitative identification, densitometry, quantitative analysis 32a

- 98 100 S. RAI, A. WAHILE, K. MUKHERJEE, B. PADA, P. MUKHERJEE\* (\*School of Natural Product Studies, Department of Pharmaceutical Technology, Jadavpur University, Kolkata 700 032, India, pknatprod@yahoo.co.in): Antioxidant activity of *Nelumbo nucifera* (sacred lotus) seeds. *J. Ethnopharmacol.* 104, 322-327 (2006). HPTLC of *Nelumbo nucifera* seed extract (standardized 50 % hydro alcoholic extract, containing 30 % of saponins, strength 10:1) on silica gel with chloroform - methanol 7:1 and hexane - ethyl acetate 7:3. Qualitative determination at 254 nm reveals six zones (at Rf 0.19, 0.36, 0.40, 0.48, 0.61 and 0.74) with the first solvent system, and nine zones (at Rf 0.10, 0.15, 0.27, 0.39, 0.42, 0.51, 0.61, 0.77 and 0.85) with the second solvent system.  
herbal, traditional medicine, HPTLC 32e
- 98 101 S. RAI, K. MUKHERJEE, M. MAL, A. WAHILE, B. SAHA, P. MUKHERJEE\* (\*School of Natural Products Studies, Department of Pharmaceutical Technology, Jadavpur University, Kolkata, India, pknatprod@yahoo.com): Determination of 6-gingerol in ginger (*Zingiber officinale*) using high-performance thin-layer chromatography. *J. Sep. Sci.* 29, 2292-2295 (2006). HPTLC of 6-gingerol in the rhizomes of *Zingiber officinale* on silica gel with n-hexane - diethyl ether 2:3. Quantitative determination by absorbance measurement. Linearity of determination of 6-gingerol is between 250 and 1200 ng and its average percentage recovery is between 99.79 - 99.84 %. The method permits a good resolution and separation of other constituents of ginger.  
quality control, herbal, densitometry, quantitative analysis, HPTLC 32g
- 98 102 E. REICH\*, V. WIDMER (\*CAMAG Laboratory, Sonnenmattstrasse 11, 4132 Muttenz, Switzerland): HPTLC for rapid identification of Black Cohosh. *LC-GC Europe* 19, July 2006 Supplement, 15 (2006). HPTLC of *Actaea racemosa* (black cohosh) extracts on silica gel with toluene - ethyl formate - formic acid 5:3:2 in the CAMAG Automatic Development Chamber ADC 2 with humidity control (saturation 20 min, humidity 5 %). Detection by dipping in sulphuric acid reagent (20 mL of sulphuric acid in 180 mL methanol) followed by heating for 5 min at 100 °C. Evaluation at 366 nm. The method is specific for identification of black cohosh and for discrimination from different species of common adulterants.  
herbal, quality control, qualitative identification, HPTLC, black cohosh 32e
- 98 103 P. RISHA, Z. MSUYA, M. NDOMONDO-SIGONDA, T. LAYLOFF\* (\*Management Sciences for Health, PO Box 50104, Dar es Salaam, Tanzania; tlayoff@msh.org): Proficiency testing as a tool to assess the performance of visual TLC quantitation estimates. *J. Assoc. Off. Anal. Chem.* 89, 1300-1304 (2006). TLC of 27 substandard drugs on silica gel with appropriate mobile phases. Routine Minilab (developed by the German Pharma Health Fund) test procedures to screen product quality and a proficiency testing program to determine the competency of health inspectors and reliability of results were established. Although the TLC screening methods provide a rapid means for drug quality assessment, they need to be put in the hands of competent users.  
quality control, qualitative identification 32a
- 98 104 D. RUDDY\*, J. SHERMA (\*Dept. of Chem., Lafayette Col., Easton, PA 18042, USA): Analysis of the caffeine in alertness tablets and caplets by high-performance thin-layer chromatography with ultraviolet absorption densitometry of fluorescence-quenched zones. *Acta Chrom.* 12, 143-150 (2002). HPTLC of caffeine in pharmaceutical preparations on silica gel with ethyl acetate -

methanol 17:3. Densitometry at 275 nm. Tablet, coated tablet, and coated caplet products containing caffeine as the active ingredient were analyzed to test the applicability of the new method.

quantitative analysis, densitometry, HPTLC, caffeine 32a

- 98 105 T. S. REDDY, A. S. REDDY, P. S. DEVI\* (\*Analytical Chemistry Division, Indian Institute of Chemical Technology, Tarnaka, Hyderabad-500007, India): Quantitative determination of sildenafil citrate in herbal medicinal formulations by high-performance thin-layer chromatography. *J. Planar Chromatogr.* 19, 427-431 (2006). HPTLC of sildenafil citrate on silica gel, pre-washed with methanol, with toluene - acetone - methanol 3:1:1 in a saturated twin-trough chamber. Quantitative determination by absorbance measurement at 312 nm. The method was validated in accordance with ICH guidelines on the validation of analytical methods.

quality control, HPTLC, densitometry, quantitative analysis 32a

- 98 106 M. SAJEWICZ\*, T. KOWALSKA (\*Inst. of Chem., Silesian Univ., 9 Szkolna Street, 40-006 Katowice, Poland): On problems with liquid chromatographic quantification of chiral 2-arylpropionic acids by use of UV-absorbance-based detection. *Acta Chrom.* 17, 292-301 (2006). TLC of ibuprofen and naproxen on silica gel F254 layer pre-washed with methanol - water 9:1 and dried at ambient temperature for 3 h. The plates were impregnated with a 0.03 M solution of L-arginine in methanol. Development with acetonitrile - methanol - water 5:1:1 containing several drops of glacial acetic acid for S-(+)-ibuprofen, and acetonitrile - methanol - water 10:2:3 containing several drops of glacial acetic acid for S-(+)-naproxen. Densitometry at 200, 205, and 210 nm for S-(+)-ibuprofen and at 202, 215, and 225 nm for S-(+)-naproxen. The investigations were performed by using three independent measurement techniques, all based on UV absorption: HPLC-UV, HPLC-DAD, and TLC-densitometry.

comparison of methods, densitometry quantitative analysis, chiral, 2-arylpropionic acids, S-(+)-ibuprofen, S-(+)-naproxen 32a

- 98 114 M. Y. SALEM\*, N. K. RAMADAN, A. A. MOUSTAFA, M. G. EL-BARDICY (\*Cairo University, Faculty of Pharmacy, Analytical Chemistry Department, Kasr-El-Aini 11562, Cairo, Egypt; maissas@hotmail.com): Stability-indicating methods for the determination of disopyramide phosphate. *J. Assoc. Off. Anal. Chem.* 89, 976-985 (2006). TLC of disopyramide phosphate on silica gel with ethyl acetate - chloroform - ammonia 17:2:1 in a pre-saturated chamber. Detection under UV light at 254 nm. Quantitative determination by absorbance measurement at 268 nm.

quality control, densitometry, quantitative analysis, disopyramide phosphate 32a

- 98 107 Breda SIMONOVSKA, Marija SRBINOSKA, Irena VOVK\* (\*National Institute of Chemistry, Laboratory for Food Chemistry, Hajdrihova 19, SI-1000 Ljubljana, Slovenia): Analysis of sucrose esters-insecticides from the surface of tobacco plant. *J. Chromatogr. A* 1127(1-2), 273-277 (2006). HPTLC of sucrose esters from the surface of leaves of *Nicotiana tabacum* L. on silica gel with n-hexane - ethyl acetate 1:3. Detection by spraying with aniline-diphenylamine reagent. Identification by off-line TLC-MS. Quantitative determination of sucrose esters-insecticides by indirect estimation through TLC of sucrose obtained after alkaline hydrolysis. Application of the method in the screening of sucrose esters in plant extracts in laboratory and field experiments.

pharmaceutical research, HPTLC, qualitative identification quantitative analysis, leaf surface compounds, *Nicotiana tabacum* L. 32e

- 98 108 B. SUHAGIA, S. SHAH, I. RATHOD, H. PATEL\*, D. SHAH, B. MAROLIA (\*Department of Quality Assurance, L.M. College of Pharmacy, Gujarat, India, patelhary2001@yahoo.com). :

Determination of gatifloxacin and ornidazole in tablet dosage forms by high-performance thin-layer chromatography. *Anal. Sci.* 22, 743-745 (2006). HPTLC of gatifloxacin and ornidazole on silica gel with n-butanol - methanol 8:1 and 1 drop ammonia with chamber saturation for 45 min. Quantitative determination by absorbance measurement at 302 nm. Linearity of determination of gatifloxacin is 100 - 500 ng and of ornidazole 250 - 1250 ng, with a correlation coefficient of more than 0.9850. The intra-day and inter-day coefficients of variation are found to be in the range of 0.68 - 2.58 % and 0.37 - 3.62%, respectively.

pharmaceutical research, quality control, HPTLC, densitometry, quantitative analysis 32a

98 109 C. SULLIVAN, J. SHERMA\* (\*Dept. Of Chem., Lafayette Coll., Easton, PA 18042-1782, USA): Determination of salicylamide in pharmaceutical tablets by high-performance thin-layer chromatography with ultraviolet absorption densitometry. *Acta Chrom.* 16, 153-163 (2006). HPTLC of salicylamide in diuretic tablets and pain-relief tablets on silica gel with concentrating zones (pre-washed with dichloromethane - acetone 1:1) with dichloromethane - acetone 4:1. Quantification by densitometry at 254 nm. The method is suitable for routine quality-control analysis of pharmaceuticals containing salicylamide.

quality control, HPTLC, quantitative analysis, densitometry, salicylamide 32a

98 082 A. K. TRIPATHI, R. K. VERMA, A. K. GUPTA, M. M. GUPTA\*, S. P. S. KHANUJA (\*Central Institute of Medicinal and Aromatic Plants, P. O. CIMAP, Lucknow-226015, India; guptammg@rediffmail.com): Quantitative determination of phyllanthin and hypophyllanthin in *Phyllanthus* species by high-performance thin layer chromatography. *Phytochem. Anal.* 17, 394-397 (2006). HPTLC of phyllanthin and hypophyllanthin from *Phyllanthus* species, e.g. *Phyllanthus amarus*, on silica gel with hexane - acetone - ethyl acetate 37:6:4. Detection by spraying with vanillin in concentrated sulfuric acid and ethanol. Quantitative determination by absorbance measurement at 580 nm. Recovery of phyllanthin is 98.7 % and of hypophyllanthin 97.3 %. The method was validated and the peak purities and limits of detection and quantification were determined.

herbal, quantitative analysis, HPTLC 32e

98 110 K. TYRPIEN\*, C. DOBOSZ, A. CHROSCIEWICZ, M. CIOLECKA, T. WIELKOSZYNSKI, B. JANSOKA, D. BODZEK (\*Dept. of Chem., Fac. of Med., Med. Univ. of Silesia, Jordana 19, 41-808 Zabrze, Poland): Investigation of nicotine transformation products by densitometric TLC and GC-MS. *Acta Chrom.* 13, 154-160 (2003). Samples of nicotine were exposed to air and UV light and the products formed (nicotyrine, cotinine, and trans-3'-hydroxycotinine) were separated by TLC on RP-18 with acetonitrile - water 22:3 in a horizontal chamber. Visual evaluation under UV 254 nm, and under white light after derivatization with Dragendorff's reagent. Quantitative determination by absorbance measurement at 260 nm. GC-MS analysis was performed to confirm the identities of the nicotine transformation products.

qualitative identification, densitometry quantitative analysis, comparison of methods, nicotine 32d

98 083 J. K. VERMA\*, A. V. JOSHI (\*Department of Chemistry, S. J. Somaiya College of Science and Commerce, Vidyavihar, Mumbai 400077, India): Rapid HPTLC method for identification and quantification of curcumin, piperine and thymol in an ayurvedic formulation. *J. Planar Chromatogr.* 19, 398-400 (2006). HPTLC of curcumin, piperine, and thymol on silica gel, pre-washed with methanol, without chamber saturation with toluene - ethyl acetate - methanol 18:2:1. Quantitative determination by absorbance measurement at 420 nm for curcumin, 333 nm for piperine, and 277 nm for thymol. Limit of detection for curcumin was 25 ng, for piperine 5 ng, and for thymol 50 ng. Rapid identification of the three compounds is also possible by spraying the plate with anisaldehyde in sulfuric acid reagent.

traditional medicine, herbal, HPTLC densitometry, qualitative identification, quantitative analysis, Ayurveda 32e

- 98 112 M. WLODARCZYK, G. MATYSIK, W. CISOWSKI, M. GLENSK\* (\*Department of Pharmacognosy, Wrocław Medical University, Nankiera 1, 50-140 Wrocław, Poland): Rapid densitometric quantitative screening of the myricitrin content of crude methanolic extracts of leaves from a variety of *Acer* species. *J. Planar Chromatogr.* 19, 378-382 (2006). TLC and HPTLC of myricitrin on silica gel, pre-washed with methanol, in a horizontal chamber with chloroform - methanol - formic acid - water 10:4:1:0.95. Quantitative determination by absorbance measurement at 254 nm.

herbal, HPTLC, quantitative analysis, densitometry 32e

- 98 113 Magdalena WÓJCIAK-KOSIOR\*, A. SKALSKA, G. MATYSIK, M. KRYSKA (\*Medical Academy, Laboratory of Planar Chromatography, Department of Chemistry, Staszica 6, 20-081, Lublin, Poland; kosiorma@wp.pl): Quantitative analysis of phenobarbital in dosage form by thin-layer chromatography combined with densitometry. *J. Assoc. Off. Anal. Chem.* 89, 995-998 (2006). HPTLC of phenobarbital on silica gel, prewashed with methanol and then acetone, using dichloromethane - ethyl acetate - formic acid 95:5:1 in a horizontal chamber. Quantitation by densitometry in the absorbance/reflectance mode at 210 nm. The validity of the HPTLC-densitometric method was established through a study of linearity, sensitivity, accuracy, and reproducibility.

quality control, densitometry, HPTLC 32a

- 98 115 M. YONAMINE\*, M. CORTEZ (\*College of Pharmaceutical Sciences, Toxicology, University of S. Paulo, Av. Professor Lineu Prestes, 580 B13B, 05509-900 Sao Paulo, SP, Brazil, yonamine@usp.br): A high-performance thin-layer chromatographic technique to screen cocaine in urine samples. *Leg. Med.* 8, 184-187 (2006). HPTLC on silica gel of cocaine urine samples submitted to solid phase extraction prior to derivatization (methylation) with diazomethane. For methylation samples were mixed with 100  $\mu$ L of a solution freshly prepared by distillation of 2.14 g N-methyl-N-nitroso-p-toluenesulfonamide with 10 mL potassium hydroxide 96 % in ethanol and 30 mL ethyl ether, and kept at room temperature for 1 min to convert benzoylecgonine to cocaine. Development over 7 cm in a saturated chamber with ethyl acetate - cyclohexane - ammonia 250:100:1. Detection by spraying with Dragendorff reagent (10 mL of 40 % m/v potassium iodide in water; 10 mL of 1 N solution of bismuth nitrate in glacial acetic acid; 80 mL of 10 % v/v sulfuric acid water solution; 2 g of resublimed iodine). The technique is capable to discriminate cocaine from interfering substances such as nicotine, caffeine and even cocaethylene in urine samples. The limit of detection was 100 ng of cocaine.

toxicology, doping, HPTLC, densitometry, quantitative analysis 32d

- 98 116 Á. Z. DÁVID, E. MINCSOVICS, I. ANTAL, É. FURDYGA, Z. ZSIGMOND, I. KLEBOVICH\* (\*Semmelweis University, Department of Pharmaceutics, Högyes Endre Street 7, 1092 Budapest, Hungary): OPLC combined with NIR spectroscopy - a novel technique for pharmaceutical analysis. *J. Planar Chromatogr.* 19, 355-360 (2006). OPLC of paracetamol, acetylsalicylic acid and caffeine on HPTLC silica gel (prewashed with acetonitrile - water 17:3) with n-hexane - toluene - diethyl ether - glacial acetic acid - methanol 100:50:30:19:1 and toluene - diethyl ether - glacial acetic acid - methanol 50:30:19:1. Quantitative determination by absorbance measurement at 254 nm.

quality control, densitometry, quantitative analysis, OPLC 32a

### 33. Inorganic substances

- 98 117 E. ADAMEK (Silesian Acad. of Med., Fac. of Pharm., Dept. of General and Anal. Chem., Jagiellonska 4, 41-200 Sosnowiec, Poland): The influence of the stationary and the mobile phases on the TLC separation of selected metal-ion complexes with sodium diethyldithiocarbamate (Na-DDTC). *Acta Chrom.* 11, 196-203 (2001). The effects of the stationary and the mobile phases on chromatographic behaviour in the separation of water-insoluble complexes of diethyldithiocarbamate with  $\text{Cd}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Mn}^{2+}$ , and  $\text{Pb}^{2+}$  have been studied by TLC with polar adsorbents (silica gel, previously activated for 40 min at 110 °C, and neutral aluminum oxide 60 Type E, activated for 4 h at 150 °C) with benzene - chloroform after chamber saturation for 2,5 h. The proportion of the components of the mobile phase was varied. Visualization by spraying with 5 % aqueous solution of  $\text{CuSO}_4$  to form a more durable yellow-green chelate (436 nm). The results obtained show that metal ions in the form of their chelates with DDTC can be separated under the presented conditions.  
qualitative identification, metal-ion complexes, sodium diethyldithiocarbamate 33a
- 98 118 M. CURTUI, Maria-Loredana SORAN\* (\*National Institute of Research and Development for Isotopic and Molecular Technology, 72-103 Donath Street, 400293 Cluj-Napoca, Romania): TLC separation of metal ions using di(n-butyl)dithiophosphoric acid and neutral organophosphorus ligands. *J. Planar Chromatogr.* 19, 297-301 (2006). TLC of U(VI), Th(IV), the lanthanides Ln(III), (La(III), Ce(III), Pr(III), Sm(III), Gd(III), Er(III)), Co(II), Ni(II), and Cu(II) on silica gel using mobile phases containing tributylphosphate, trioctylphosphine oxide (TOPO), and di(n-butyl)dithiophosphoric acid (HDBDTP). The results obtained showed that the retention factors of U(VI), Th(IV), and Ln(III) are enhanced when a mixture of HDBDTP and TOPO is used. TLC on silica gel with xylene - ethyl methyl ketone - N,N-dimethylformamide 16:2:1 in an unsaturated chamber. Detection with 0.05 % arsenazo(II) for U(VI), Th(IV), and Ln(III) and with 0.1 % rubeanic acid in ethanol for Co(II), Ni(II), and Cu(II).  
33a
- 98 119 V. GHOULIPOUR, S. W. HUSAIN\* (\*Fac. of Chem., Univ. of Tarbiat Moallem, 49, Mofatteh Avenue, Tehran-15614, Iran): Inorganic ion-exchangers for quantitative TLC of toxic elements. V. Separation and determination of chromium (VI). *Acta Chrom.* 12, 170-176 (2002). Rapid and selective method for the separation and determination of Cr (VI) from Al (III), Cr (III), Mn (II), Fe (III), Co (II), Ni (II), Cu (II), Zn (II), Se (IV), Sr (II), Zr (IV), Cd (II), La (III), Ce(III), Hg(II), and U (VI). TLC on self-prepared titanite silicate ion-exchange plates with 0.4 M ammonium oxalate - 2.2 M aqueous ammonia 1:1. Detection of Cr (VI) by spraying with a saturated solution of diphenyl-carbazide in iso-propanol. Densitometry at 545 nm.  
toxicology, quantitative analysis, toxic cations 33a
- 98 120 A. MOHAMMAD\*, M. AJMAL, S. ANWAR (\*Dept. of Appl. Chem., Z. H. College of Engineering and Technology, Aligarh Muslim Univ., Aligarh-202 002, India): Identification of  $\text{Hg}^{2+}$  and its separation from  $\text{UO}_2^{2+}$  and  $\text{Fe}^{3+}$  on impregnated silica gel layers with formic acid - sodium chloride mobile phases. *Acta Chrom.* 9, 113-122 (1999). Selective separation of  $\text{Hg}^{2+}$  from spiked sea and industrial waste water samples. TLC of some inorganic pollutants on silica gel impregnated with potassium thiocyanate (0.01 - 2.0 M) 1:1 mixtures of sodium chloride (0.01 - 1.0 M) and formic acid (0.01 - 1.0 M) as mobile phases. Detection with 5 % potassium ferrocyanide in water (for  $\text{Fe}^{3+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{VO}_2^{+}$ , and  $\text{Tl}^{+}$ ), 0.05 % dithizone in carbon tetrachloride (for  $\text{Zn}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Bi}^{3+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Ag}^{+}$ , and  $\text{Tl}^{+}$ ), 1 % dimethylglyoxime in ethanol (for  $\text{Ni}^{2+}$  and  $\text{Co}^{2+}$ ), 0.1 % aluminum (for  $\text{Al}^{3+}$ ) and 0.01 % thorine (for  $\text{Th}^{4+}$  and  $\text{Zr}^{4+}$ ). The effect of surfactants, halides and oxyanions on the separation of  $\text{Hg}^{2+}$  from  $\text{UO}_2^{2+}$  and  $\text{Fe}^{3+}$  was examined.

The method was used to identify  $\text{Hg}^{2+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Cd}^{2+}$ , and  $\text{Zn}^{2+}$  in synthetic heavy metal sludge samples.

environmental, qualitative identification, inorganic pollutants 33a

- 98 121 A. MOHAMMAD\*, M. P. A. NAJAR (\*Anal. Res. Lab., Dept. of Appl. Chem., Z.H. Col. of Eng. and Techn., Aligarh Muslim Univ., Aligarh-202002, India): Quaternary separation of some transition metal chlorosulphates on mixed adsorbent layers with water as mobile phase. Quantitative determination of nickel chlorosulphate. *Acta Chrom.* 11, 154-170 (2001). Several stationary phase - mobile phase combinations were evaluated to identify suitable chromatographic systems for analysis of metal chlorosulphates. The effect of anionic species on the mobility and separation of metal chlorosulphates was examined. Quaternary separation of the chlorosulphates (Mn, Fe, Ni, and Cu or Zn) on silica gel - cellulose 2:1 with double-distilled water. Detection of Cu and Fe by spraying with 1 % aqueous potassium ferrocyanide, of Ni and Co with 1 %, and of Zn and Mn with 0.5 % dithizone in chloroform. For the quantitative determination of nickel the substances were extracted with distilled water, and buffer solution (1 M ammonium hydroxide - 1 M ammonium chloride 1:1) with bromo-pyrogallol red indicator (0.05 g in 100 mL ethanol - water 1:1) was added. The mixture was titrated with 0.05 M aqueous EDTA solution.

metal cations 33a

- 98 122 A. MOHAMMAD\*, N. JABEEN (\*Anal. Lab., Dept. of Appl. Chem., Zakir Hussain Col. of Eng. and Techn., Aligarh Muslim Univ., Aligarh-202 002, India): Reversed-phase chromatography of amines, phenols, and metal cations on silica layers impregnated with tributyl phosphate, using surfactant-mediated mobile phases. *Acta Chrom.* 13, 135-153 (2003). TLC of 17 metal cations (e.g.  $\text{Cr}^{3+}$  from  $\text{Cr}^{6+}$ ,  $\text{Fe}^{3+}$  from  $\text{Mn}^{2+}$  and  $\text{Cr}^{6+}$ ,  $\text{VO}^{2+}$  from  $\text{Mn}^{2+}$  and  $\text{Cr}^{6+}$ ) and 14 phenol derivatives (e.g. o-cresol from m-cresol, m-aminophenol from o-aminophenol) on silica gel impregnated with 0.001 M tributyl phosphate, with an 0.01 M aqueous micellar solution of N,N,N-cetyltrimethyl ammonium bromide as mobile phase. An aqueous solution ( $8.3 \times 10^{-6}$  M) of the non-ionic surfactant, Brij-35 proved suitable for achieving good separations of 16 amines (e.g. p-dimethylaminobenzaldehyde from L-tryptophan, p-dimethylaminobenzaldehyde from indole) on silica layers impregnated with 0.001 M TBP. Visualization of amine or phenol spots by treatment with iodine vapour for 10 min. Metal ions were detected by spraying with the appropriate chromogenic reagent.

qualitative identification, metal cations, phenol derivatives, amines 33a, 7

- 98 123 A. MOHAMMAD\*, R. YOUSUF, Y. HAMID (\*Anal. Res. Lab., Dept. of Appl. Chem., Fac. of Eng. and Techn., AMU, Aligarh-202002, India): Thin layer chromatography of inorganic ions on blended inorganic ion-exchangers with tributyl phosphate - formic acid as mobile phase. *Acta Chrom.* 11, 171-182 (2001). The chromatographic behaviour of some cations ( $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{VO}_2^{2+}$ ,  $\text{Fe}^{2+/3+}$ ,  $\text{Al}^{3+}$ ,  $\text{Th}^{4+}$ ,  $\text{Mo}^{6+}$ ,  $\text{W}^{6+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Hg}^{+2+}$ ,  $\text{Bi}^{3+}$ ,  $\text{Ag}^{+}$ , and  $\text{Tl}^{+}$ ) and anions ( $\text{CrO}_4^{2-}$ ,  $\text{Cr}_2\text{O}_7^{2-}$ ,  $\text{IO}_3^-$ ,  $\text{IO}_4^-$ ,  $\text{BrO}_3^-$ ,  $\text{SCN}^-$ ,  $\text{Fe}(\text{CN})_6^{3-}$ ,  $\text{NO}_2^-$ ) was examined on layers prepared from 1:9 mixtures of a synthetic inorganic ion-exchanger (stannic arsenate or tin(IV) molybdosilicate) with silica gel, alumina, or cellulose with 1 % methanolic tri-n-butyl phosphate - 1 M aqueous formic acid 1:4 as mobile phase. Several binary cation separations of analytical interest were achieved. The metal ions were detected by use of conventional spot-test reagents. Separation of  $\text{IO}_3^-$  from  $\text{NO}_2^-$  and  $\text{BrO}_3^-$  under the same conditions, detection with 0.2 - 0.5 % diphenylamine in 2 M  $\text{H}_2\text{SO}_4$ . For quantitative determination the region corresponding to  $\text{IO}_3^-$  was isolated from the plate and extracted with 1.0 M hydrochloric acid. 1 % potassium iodide and conc. hydrochloric acid were added to the extract and the mixture obtained was titrated against  $\text{Na}_2\text{S}_2\text{O}_3$ .

qualitative identification, metal cations, anions 33a

- 98 124 A. MOHAMMAD\*, S. SYED, L. M. SHARMA, A. A. SYED (\*Dept. of Appl. Chem., Fac. of Eng. and Techn. A.M.U., Aligarh-202002, India): Thin layer chromatographic separation and recovery of gold and silver from secondary sources. *Acta Chrom.* 11, 183-195 (2001). TLC method for selective separation of Au<sup>3+</sup> and Ag<sup>+</sup> from accompanying metal ions (Cr<sup>6+</sup>, Ni<sup>2+</sup>, Cu<sup>2+</sup>, Zn<sup>2+</sup>, Cd<sup>2+</sup>, and Hg<sup>2+</sup>). TLC of Au<sup>3+</sup> on silica gel G with aqueous 1.2 mM N-cetyl-N,N,N-trimethylammonium bromide and of Ag<sup>+</sup> on alumina G layers with 2.5 M aqueous ammonium sulphate. Detection of Au<sup>3+</sup> by spraying with 5 % aqueous stannous chloride, of Ag<sup>+</sup>, Cd<sup>2+</sup>, and Hg<sup>2+</sup> by spraying with yellow ammonium sulphide, and of Cu<sup>2+</sup> by spraying with 1 % aqueous potassium ferrocyanide. Ni<sup>2+</sup> was detected by spraying with 1 % ethanolic dimethylglyoxime, Cr<sup>6+</sup> by spraying with saturated ammonium thiocyanate in acetone, and Zn<sup>2+</sup> by spraying with 0.05 % dithizone in CCl<sub>4</sub>. TLC in combination with spectrophotometry and volumetric analysis was used to determine content of Au<sup>3+</sup> and Ag<sup>+</sup> in secondary materials.
- qualitative identification, cations 33a
- 98 125 A. MOHAMMAD\*, V. AGRAWAL (\*Anal. Research Lab., Dept. of Applied Chem., Fac. of Engineering and Technology, Aligarh Muslim Univ., Aligarh 202002, India): Use of cationic micellar mobile phases in normal-phase TLC for enhanced selectivity in the separation of transition metal ions. Simultaneous separation of mixtures of zinc, nickel, mercury and cadmium or manganese cations. *Acta Chrom.* 12, 177-188 (2002). The effect of surfactant concentrations below and above the critical micellar concentration on the retention behaviour of metal ions was examined. The effect of organic and inorganic additives on the mobility and separation efficiency of the metal ions was also assessed. TLC of mixtures of Zn<sup>2+</sup>, Ni<sup>2+</sup>, Hg<sup>2+</sup>, and Cd<sup>2+</sup> or Mn<sup>2+</sup> on silica gel with 50 mM aqueous micellar solution of N,N-cetyltrimethyl-ammonium bromide (CTAB). Detection of Ni<sup>2+</sup> with a 1 % solution of alcoholic dimethylglyoxime, of Zn<sup>2+</sup>, Cd<sup>2+</sup>, and Hg<sup>2+</sup> with a 0.5 % solution of dithizone in CCl<sub>4</sub>, and of Mn<sup>2+</sup> with a 1:1 mixture of 2 M sodium hydroxide and 30 % H<sub>2</sub>O<sub>2</sub>. The proposed method was used for separation and identification of cations from drag-out nickel-plating solution and a sludge sample containing their hydroxides, as well as for semi-quantitative Ni<sup>2+</sup>-determination.
- qualitative identification, micellar TLC, cations 33a
- 98 126 A. MOHAMMAD\*, Y. H. SIRWAL (\*Anal. Lab., Dept. of Applied Chem., Fac. of Engineering and Technology, Aligarh Muslim Univ., Aligarh-202002, India): Novel mobile phase for separation of Cr<sup>6+</sup> from Cr<sup>3+</sup> and associated heavy metal cations by high-performance thin-layer chromatography. *Acta Chrom.* 13, 117-134 (2003). HPTLC of 11 heavy metal cations on silica gel with pure organic, mixed organic and mixed aqueous - organic mobile phases. Mobile phases such as methanol - dimethylamine 4:1 and methanol - dimethylamine - formic acid 4:4:1 were found most suitable for rapid separation and identification of mixtures of Cr<sup>6+</sup> and Cr<sup>3+</sup> and of Cr<sup>6+</sup>, Ni<sup>2+</sup> and Co<sup>2+</sup>, respectively. Detection of Cd<sup>2+</sup>, Ag<sup>+</sup>, Pb<sup>2+</sup>, Tl<sup>+</sup>, Bi<sup>3+</sup> and Hg<sup>2+</sup> by spraying with yellow ammonium sulphide reagent, of VO<sub>2</sub><sup>+</sup> with a 1 % aqueous solution of potassium ferrocyanide, of Ni<sup>2+</sup> and Co<sup>2+</sup> with dimethylglyoxime (0.2 % in ammonia), of Cr<sup>6+</sup> with a saturated alcoholic solution of AgNO<sub>3</sub> and of Cr<sup>3+</sup> with a 1 % alizarin red in methanol. The effect of impurities such as inorganic ions, phenols, and surfactants on the separation of Cr<sup>6+</sup> and Cr<sup>3+</sup> was examined. The proposed method was successfully used for analysis of industrial wastewater samples.
- environmental, qualitative identification HPTLC, metal cations 33a

### 35. Other technical products and complex mixtures

- 98 127 C. IMARK, M. KNEUBUEHL, S. BODMER\* (\*Biodyn GmbH, Industriestrasse 31, CH-8305 Dietlikon, Switzerland, bodmer@biodyn.ch) : Occurrence and activity of natural antioxidants in herbal spirits. *Innovative Food Science and Emerging Technologies* 1, 239-243 (2001). HPTLC



of commercial herbal spirits (alcoholic or hydroalcoholic solutions of volatile substances with flavoring or medicinal properties) and one red wine on silica gel with toluene - ethyl formate - formic acid 79:20:1. Antioxidative components were detected by dipping for 30 s in a soybean oil solution (3 % in n-hexane, previously treated with active carbon). Quantitative determination in UV light at 254 nm after different times of UV-exposure (30 min - 20 h). The antioxidant activity could be evaluated from the fluorescence-persisting time of the respective spots and was correlated with linoleic acid oxidation and DPPH-titration methods. Although the nature of the active herbal antioxidants remains to be established, phenolic compounds seem to be key candidates.

food analysis, herbal, HPTLC, quantitative analysis, densitometry, comparison of methods, post-chromatographic derivatization 35b

- 98 128 A. MOHAMMAD\*, H. SHAHAB (\*Anal. Res. Lab., Dept. of Applied Chem., Fac. of Eng. and Tech., Aligarh Muslim Univ., Aligarh-202002, India): Use of a glutamic acid-containing aqueous-organic mobile phase for on-plate separation, detection, and identification of cationic and non-ionic surfactants by thin-layer chromatography. *Acta Chrom.* 17, 272-291 (2006). TLC on silica gel with 0.1 M glutamic acid - methanol - acetone 1:1:1 has been found to be highly suitable for separation and identification of cationic and non-ionic surfactants. Visualization by use of Dragendorff reagent or iodine vapour. Spectrophotometric determination of tetradecyltrimethylammonium bromide at 670 nm after the spot extraction. The method has been used for identification of tetradecyltrimethylammonium bromide and Triton TX-100 in saline water, river water, and domestic waste water. The effects of sample pH, polarity of the alcohol and nature of the amino acid in the mobile phase, and the presence of alumina, kieselguhr, or cellulose in the silica gel layer have been examined.

qualitative identification, cationic and non-ionic surfactants 35a

- 98 129 T. WIDLA\*, M. SLIWIOK (\*Fac. of Law and Admin., Dept. of Criminalistics, Silesian Univ., 40-006 Katowice, Bankowa Street, Poland): Detection and determination of trotyl by HPTLC. *Acta Chrom.* 6, 113-115 (1996). HPTLC of trotyl on silica gel (activated at 110 °C) with hexane - benzene 1:1. Several selected visualizing agents were investigated: phenol red, bromphenol blue, thymol blue and bromothymol blue. After spraying, the plates were heated at 100 °C. A low detection limit (1 µg) was obtained by application of phenol red and bromphenol blue. This method enables further possibilities for quantitative determination.

HPTLC, qualitative identification, forensic application, trotyl 35

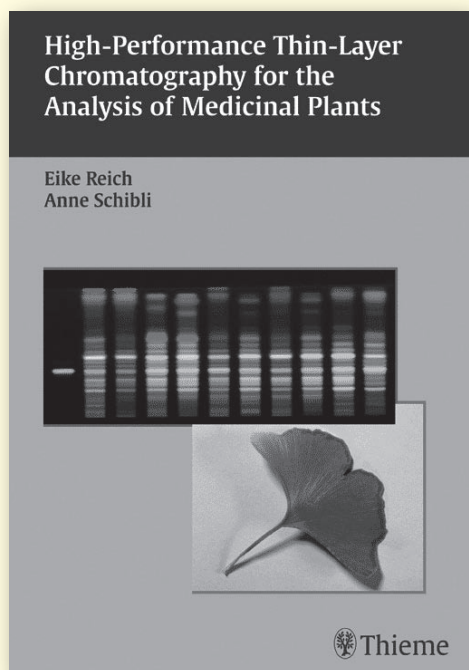
### 37. Environmental analysis

- 98 130 D. BODZEK\*, C. DOBOSZ, K. TYRPIEN (\*Dept. of Chem., Fac. of Med., Med. Univ. of Silesia, 41-808 Zabrze, Jordana 19, Poland): Determination of selected PAH carbonyl derivatives by TLC with densitometric detection. *Acta Chrom.* 11, 108-117 (2001). TLC of polyaromatic carbonyl compounds most commonly found in environmental samples (acridone, 1,2-naphthoquinone, 9,10-phenanthrenequinone, acenaphthenequinone, xanthone, 1-aminoanthraquinone, anthrone, 1,4-chrysenoquinone, anthraquinone and 9-fluorenone) on silica gel and RP-18. Different combinations of solvents were evaluated as mobile phases. The best separation was obtained by use of pure dichloromethane on silica gel and methanol - water - acetonitrile 3:2:1 on RP-18 in a horizontal chamber. Evaluation under UV 254 nm and 366 nm. Quantitative determination of acridone by fluorescence measurement at 390/>400 nm, and by absorbance measurement at 250 nm for the other compounds.

environmental, densitometry, quantitative analysis, PAH, carbonyl derivatives 37c

**38. Chiral separation**

- 98 131 M. SAJEWICZ, H. E. HAUCK, G. DRABIK, E. NAMYSLO, B. GLÓD, Teresa KOWALSKA\* (\*Institute of Chemistry, Silesian University, 9 Szkolna Street, 40-006 Katowice, Poland): Tracing possible structural asymmetry of silica gel used for precoating thin-layer chromatography plates. *J. Planar Chromatogr.* 19, 278-281 (2006). Discussion of the deviation from the strict vertical of the migration tracks of chiral compounds during TLC on silica gel. Focus is put on possible crystalline asymmetry of the silica gel used in TLC. Description of an attempt to verify the hypothesis by means of circular dichroism spectroscopy. TLC of S-(+)-ibuprofen and S-(+)-naproxen on silica gel with acetonitrile - methanol - water 5:1:1 for S-(+)-ibuprofen and 10:2:3 for S-(+)-naproxen. Quantitative determination by absorbance measurement at 210 nm.  
quantitative analysis, densitometry 38
- 98 132 M. SAJEWICZ, R. PIETKA, G. DRABIK, E. NAMYSLO, Teresa KOWALSKA\* (\*Institute of Chemistry, Silesian University, 9 Szkolna Street, 40-006 Katowice, Poland): On the stereochemically peculiar two-dimensional separation of 2-arylpropionic acids by chiral TLC. *J. Planar Chromatogr.* 19, 273-277 (2006). TLC of the 2-arylpropionic acids S-(+)-ibuprofen, S,R-(+/-)-ibuprofen, S-(+)-naproxen, and S,R-(+/-)-2-phenylpropionic acid on silica gel pre-developed with methanol - water 9:1 and impregnated with L-arginine in the cationic form as chiral ion-pairing reagent. Acetonitrile - methanol - water 5:1:1 (with several drops of glacial acetic acid to fix the pH at < 4.8) was used for ibuprofen, 10:2:3 for naproxen and 20:4:3 for 2-phenylpropionic acid. Quantitative determination by absorbance measurement at 210 nm.  
qualitative identification, densitometry 38
- 98 133 J. SZULILK\*, A. SOWA (\*Inst. of Chem., Silesian Univ., 9 Szkolna Street, 40-006 Katowice, Poland): Separation of selected enantiomers of fatty hydroxyl acids by TLC. *Acta Chrom.* 11, 233-237 (2001). Conditions for separation of racemic mixtures of fatty acids by use of a mobile phase containing a chiral compound as additive and on a chiral stationary phase. TLC separation of the enantiomers of DL-alpha-hydroxypalmitic acid on silica gel with acetone - hexane - hydrochloric acid 3:5:2 containing 1 % L-alanine, detection by dipping in 2 % aqueous sodium hydroxide solution. TLC of the enantiomers of DL-12-hydroxystearic acid on silica gel F254 with hexane - acetonitrile - acetic acid - hydrochloric acid 5:2:1:2 containing 2 % L-alanine, detection by dipping in 2 % aqueous sodium hydroxide solution. Separation of the enantiomers of DL-12-hydroxyoleic acid on chiral plate with acetone - water 3:2, detection by treatment with iodine vapor. Isomers were characterized by calculation of their optical topological indexes.  
qualitative identification, fatty acids, enantiomers 38



E. Reich, A. Schibli

## High-Performance Thin-Layer Chromatography for the Analysis of Medicinal Plants

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264 pages, 175 photos and graphs,  
43 references

### Key areas

1. Medical Plants-Analysis
2. Herbs-Analysis
3. Thin-Layer Chromatography
4. Plant Extracts-Analysis
5. Plants, Medicinal-Chemistry
6. Chromatography

This is a book one can fall in love with it reading it from page to page. Well somebody will skip theory. But all practical aspects are dealt with completely, carefully, easy to understand and up to the level where it could be reproduced in practice by the reader.

The book as such is of high production quality cleanly printed on top quality paper throughout the whole volume.

Colour is intensively used for a top level of organization, a truly great help for the normally information overloaded reader. The HPTLC graphics are extremely well reproduced with respect to their original life colour. All technical drawings are excellent done and organized. They contain the messages in a high level of communication to the reader. In summary to what one sees at first: a book of first class, rarely seen elsewhere.

### Quantitative data

Introduction - short but concise	7 % of the total content
Theoretical concepts	15 %
Practical aspects of modern TLC	23 %
Typical applications in herbal analysis	12 %
Method development	2 %
Validation	8 %
Appendices in sum	6 %
<small>Definitions - Standard operating procedure for HPTLC of herbal raw materials - Resources - Common mobile phases for the screening of unknown herbal drugs - most common non specific derivation reagents for analysis of herbal drugs - Worksheet - Validation of method for identification of eleuthero by HPTLC fingerprint.</small>	
Index	0.4 % (5 pages).

### **Some qualitative data – analyzing the content for applicability**

The authors avoided to show the beauty of the analysis of Medicinal Plants only, although just the cover page picture invites analysts to immediately start with HPTLC. On page 181 he/she may see what may be ahead of his/heir work in case one is confronted with “unknown botanicals”. Under “Common Mobile Phases for the Screening of Unknown Herbal drugs” the reader finds more than 25 precisely and quantitatively given mobile phase mixtures for 10 substance classes which may help him to succeed in an own case. On page 185 he finds a table of miscibility of various mobile phases and sample solvents he might not have in easy reach in his library. Under “Most common Non specific Derivatization Reagents for Analysis of Herbal Drugs” the reader finds four full pages in tabular order “Reagent name/Preparation, use/Examination/Detection of”. Again, there is a precise quantitative receipt what to mix, what to check for, what to avoid, how to use. Thus there is immediate help again not so easy to find elsewhere. Reichs/Schiblis book can really be used the classical way: to read, to understand, to do. There is no electronic full text search and text analysis necessary. This book is made for being kept open at the working place.

However it is also highly qualified to inform, teach and change minds. Analysts who know HPLC only should fear to inject any plant extract into their expensive column, especially in the case of HPLC x HPLC for comprehensive separations (which they think it will solve all problems). These HPLC-only analysts may dramatically change their mind. Thus this book is dangerous for them.

Professor Dr. Rudolf E. Kaiser

Institute for Chromatography  
Bad Dürkheim

## Chlorine-free mobile phase for determination of PAH in water extracts



13

▲ Dr. Holger Hegewald

Dr. Hegewald\* is general manager of an analytical laboratory in Évora, Portugal. For analysis of pesticides, plant hormones, aflatoxin M1 in milk, anthocyanes and organic acids in wine as well as polycyclic aromatic hydrocarbons (PAH) in water he exclusively uses planar chromatography. In CBS 95 he has reported about the time-efficient derivatization of glyphosate and AMPA at the starting zone – here he is presenting a chlorine-free mobile phase for the determination of PAH.

### Introduction

The method is based on the German standard DIN 38407-7 [1] for quantitative determination of 6 PAH on caffeine-impregnated HPTLC plates with dichloromethane at  $-20^{\circ}\text{C}$ .

**The main issue of mobile phase selection should primarily be selectivity to obtain the best separation performance. However, also health and environmental aspects should be considered. Generally the substitution of carcinogenic and mutagenic benzene by toluene is recommended, both belong to the same selectivity group according to Snyder [2]. However, chlorinated solvents of possible carcinogenic risks, besides environmental ones, are forcing to be substituted because e. g. chloroform is the single representative of the selectivity group VIII, besides water, and with respect to dichloromethane not any chlorine-free solvents of the same selectivity group (V) exists at all.**

Using isopropyl acetate as chlorine-free solvent instead of dichloromethane allowed high-performance separations showing good base line characteristics and advantageous matrix retardation – all

in all perfectly suited for quantitative evaluation of chromatograms.

Qualitative determination of PAH for screening was performed in the horizontal developing chamber with isopropyl acetate – n-hexan 3:1 at room temperature.

### Sample preparation

Water samples were extracted by solid phase extraction on C18 sorbents, followed by elution of the adsorbed PAH with 3 mL dichloromethane. After addition of the internal standard (IS) 2-methyl anthracene (2-MA) the extracts were concentrated with nitrogen to about 0.1 mL.

### Standard solution

Dilution of the certified standard solutions (10 ng/ $\mu\text{L}$ ) with methanol to obtain the following standard mixtures:

Substance	Concentration (ng/ $\mu\text{L}$ )
Benzo(a)pyrene	0.1
Benzo(ghi)perylene	0.1
Benzo(b)fluoranthene	0.2
Benzo(k)fluoranthene	0.2
Indeno(1,2,3-cd)pyrene	0.2
Fluoranthene	0.5
2-Methyl anthracene (IS)	1.5

### Layer

HPTLC plate silica gel 60 F<sub>254</sub> caffeine-impregnated for PAH determination (Merck) or Nano-SIL-PAH (Macherey-Nagel), 20 x 10 cm.

### Sample application

Bandwise with Linomat, band length 7 mm, 17 tracks, application volume 1–12  $\mu\text{L}$  of the standard solution and about one third of the sample extracts, track distance 10 mm, distance from the side and lower edge 20 and 8 mm, respectively, application rate 7 s/ $\mu\text{L}$ . After application the start zones were dried for 2 min in a stream of cold air.

For screening in the horizontal developing chamber 34 standards and sample extracts can be applied, also on the opposite side, for parallel development from both plate sides.

## Chromatography

**Quantitative HPTLC:** In the precooled ( $-20\text{ }^{\circ}\text{C}$ , 30 min) twin trough chamber with 8 mL isopropyl acetate in one trough without chamber saturation. After application the dry plate was first equilibrated in the solvent-free trough for 10 min at  $-20\text{ }^{\circ}\text{C}$ . Chromatography started by placing the plate into the solvent-containing trough. Migration distance from the lower plate edge was 70 mm (migration time 25 min).

**Qualitative HPTLC:** At room temperature in the horizontal developing chamber with isopropyl acetate – *n*-hexane 3:1 (v/v) if desired from both sides, S-configuration. Migration distance from the lower plate edge was 50 mm (migration time 9 min).

**For both:** After drying (2 min in a stream of cold air) the plate was immersed in paraffin – toluene 1:1 for fluorescence intensification with the TLC plate immersion device III (rate 4 cm/s, time 1 s).

## Densitometric evaluation

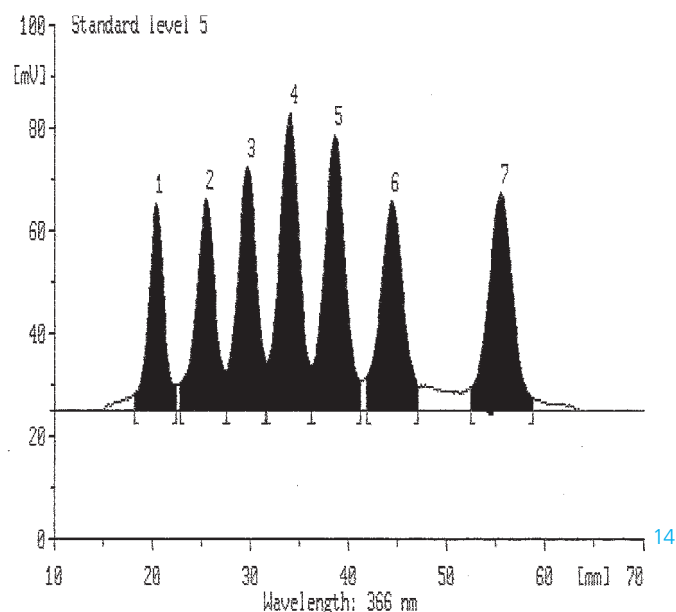
TLC scanner with CATS software, fluorescence measurement at UV 366/ $>400$ , linear calibration by peak height.

## Documentation

With DigiStore2 documentation system by illumination at UV 366/ $>400$  nm.

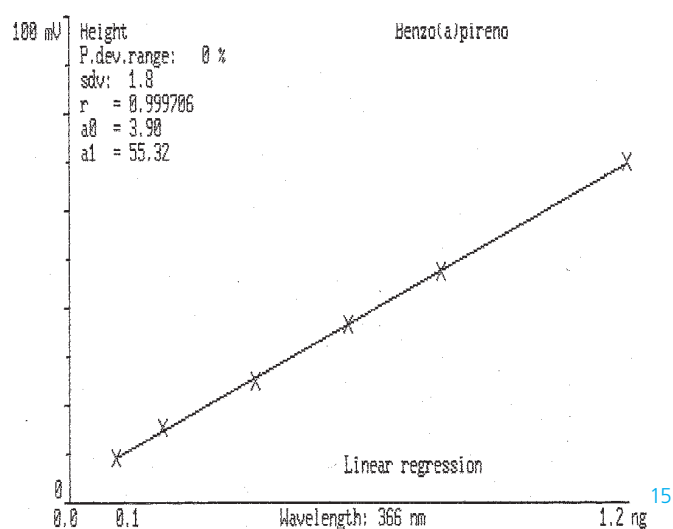
## Results and discussion

**Quantitative HPTLC:** The following densitogram shows the separation of a 10  $\mu\text{L}$ -standard mixture. The elution order of the PAH was the same as for the dichloromethane separation according to DIN 38407-7 and of comparable separation quality.



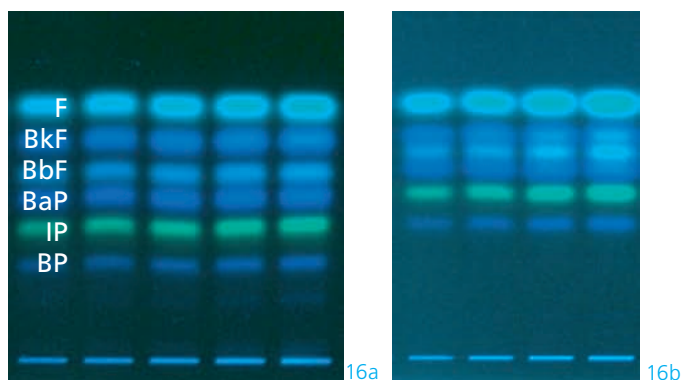
▲ Quantitative HPTLC: Densitogram of a standard track (1–5 ng/band, IS 15 ng/band): 1 benzo(ghi)perylene (BP), 2 indeno(1,2,3-cd)pyrene (IP), 3 benzo(a)pyrene (BaP), 4 benzo(b)fluoranthene (BbF), 5 benzo(k)fluoranthene (BkF), 6 fluoranthene (F), 7 2-methyl anthracene (2-MA) as IS

The calibration shown for BaP ranged from 0.1 to 1.2 ng/band is linear with a relative standard deviation of  $\pm 1.8\%$  ( $r > 0.9997$ ).



▲ Linear calibration of BaP in the range of 0.1–1.2 ng/band

Both presented chromatographic systems seem to be more robust regarding matrix interferences as the separation with dichloromethane.



▲ Quantitative (left) and qualitative HPTLC (right): Plate section with standard tracks (2–12 ng/band, in this case higher amounts were used for visualization) under illumination at UV 366 nm (here without peak 7, the internal standard 2-methyl anthracene)

Moreover, the qualitative system with isopropyl acetate – *n*-hexane had the advantage that all 6 PAH were sufficiently well separated for visual evaluation under UV 366 nm whereas only up to 5 bands were visible with the solvent system according to DIN 38407-7.

Based on the selective fluorescence detection, differences in the densitogram quality were not observed whether plate pre-washing was employed or not. Thus pre-washing of the plates was skipped.

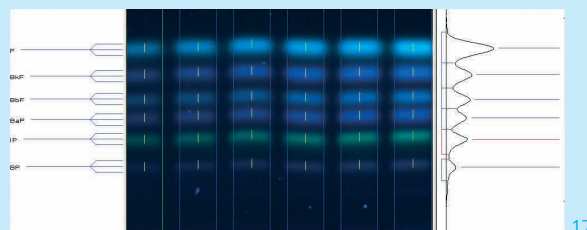
*Note of the editor: Using isopentyl acetate instead of isopropyl acetate causes banana smell in the laboratory.*

Further information is available from the author on request.

\*Dr. Holger Hegewald, Lacrome Lda, Rua César Batista 6 D, P-7000-715 Évora, Portugal, lacrome@clix.pt

[1] German Standard DIN 38407-7: Determination of six polycyclic aromatic hydrocarbons by HPTLC, Beuth Verlag, Berlin 2000.

[2] Snyder. L.R., J. Chromatogr. Sci. 16, 223-234,1978



### VideoScan Digital Image Evaluation

For PAH screening the plate can visually be evaluated with the DigiStore 2 documentation system using a high resolution 12 bit CCD camera. For rapid quantification of the water samples showing PAH findings around the limit value the digital image evaluation software VideoScan can be used. VideoScan complies with GMP/GLP and can be IQ/OQ qualified. Flexible features such as profile comparison of tracks from several chromatograms and evaluation of tracks with variable distance or distorted tracks are available. Chromatograms can be evaluated at any time, even years after capture.

### Key features of the VideoScan program at a glance

- Rapid and easy to use
- Integration of the analog curves can be performed automatically or manually.
- Quantitative evaluation can be performed via peak area and/or peak height.
- Single or multi level calibration (linear or polynomial regression) can be selected.

## Evidence of the metabolites desphenyl-chloridazon and methyl-desphenyl-chloridazon in surface, ground and drinking water



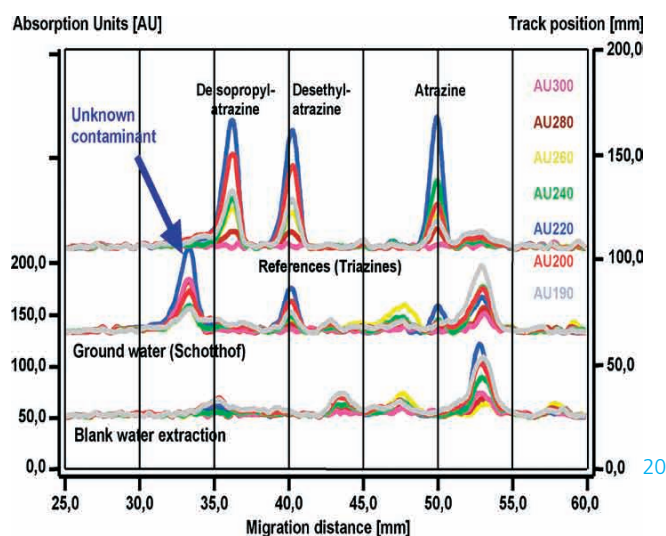
▲ Dr. Walter H. Weber\* 18



▲ Dipl.-Ing. (FH) Wolfram Seitz,  
Dr. Wolfgang Schulz, Anna Aichinger 19

At the laboratory for operation control and research of Zweckverband Landeswasserversorgung (LW), which was previously introduced in CBS 94 and 95, respectively, two formerly unknown water contaminants were identified. During the routine investigation of the water resources within the scope of "multidimensional screening" HPTLC/AMD-UV/VIS was applied and two unknown contaminants were found. For structure elucidation the detected substance zones were transferred by the extraction device ChromeXtract® (CBS 93, 94 and 96) to tandem-mass spectrometry and time-of-flight mass spectrometry. Thus, the corresponding protonated molecules were detected and evidence of the presence of the metabolites desphenyl-chloridazon and methyl-desphenyl-chloridazon were made by HPTLC/MS [1].

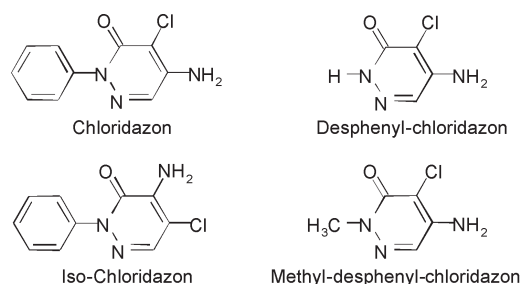
**The philosophy of the multidimensional screening concept is to apply biological test systems for the assessment of the toxicity of a sample or the usually unknown contaminants in addition to target analysis (GC and HPLC). Here, HPTLC with automated multiple development (AMD) is of particular importance to screen for a wide polarity range regarding toxic substances present in the sample. For identification of toxic unknowns HPTLC/MS coupling is an essential tool.**



▲ AMD monitoring (recorded by multi-wavelength scan) within the scope of "multidimensional screening"

### Introduction

The agent chloridazon is a systemic herbicide approved for the application in Germany, which is used for the control of weeds in the cultivation of sugar-beets, fodder beet, and beetroot. Desphenyl-chloridazon is the primary metabolite of this herbicide and up to now was not included in monitoring programs of surface water, ground water, or drinking water. Also, methods for the determination of the metabolites were previously not published. Therefore, the present work describes a method for the detection of chloridazon and its metabolites using HPTLC/AMD.



▲ Structure formulae of the herbicide chloridazon, its metabolites desphenyl-chloridazon and methyl-desphenyl-chloridazon as well as iso-chloridazon



## Sample preparation

Solid phase extraction (SPE) was used for analyte enrichment: 1 L water sample was adjusted to pH 3 with sulfuric acid and sucked through the SPE sorbent. Cleaning and conditioning of the SPE cartridge (200 mg sorbent, e.g. SDB1, Mallinckrodt-Baker or Isolute ENV+, Separtis) was performed with 6 mL *n*-hexan, acetone and methanol each and twice with 6 mL MilliQ water (pH 3). For elution 4 mL methanol was used. After concentration in a gentle stream of nitrogen the residue was taken up in 200 µL methanol.

For waters with high matrix-load (e.g. for some surface waters) a further purification step, like gel permeation chromatography (GPC), is recommended.

## Standard solution

Chloridazon, desphenyl-chloridazon, iso-chloridazon (Dr. Ehrenstorfer) and methyl-desphenyl-chloridazon (BASF) were dissolved in methanol (100 mg/L each) and stored at -20 °C.

## Layer

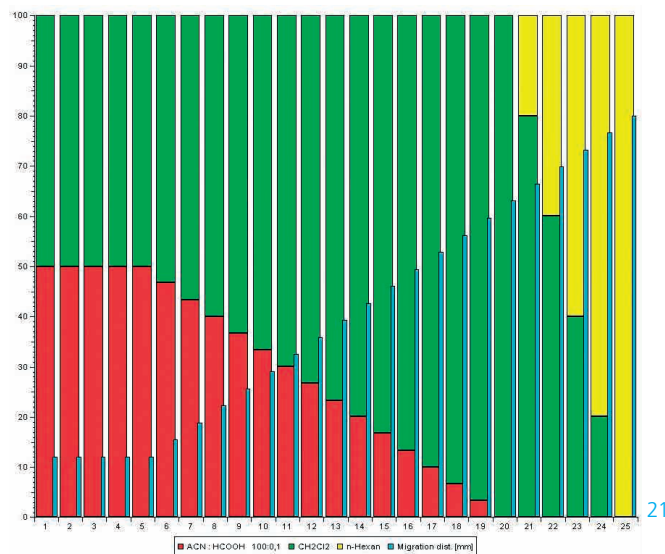
HPTLC plate silica gel 60 F<sub>254</sub> (Merck), 20 × 10 cm

## Sample application

With Automatic TLC Sampler 4 (ATS4), 16 tracks, band length 6 mm, track distance 10 mm, distance from lower plate edge 8 mm, distance from both sides at least 20 mm

## Chromatography

In the AMD 2 system with a 25-step gradient based on dichloromethan. To increase the elution power in the first steps acetonitrile – formic acid 100:0.1 (v/v) is added at the beginning. To decrease the elution power later on, *n*-hexane is added. The migration distance of the first 5 steps is set to 12 mm, whereas the migration distance of the following 21 steps is equidistant prolonged by 3.4 mm. The last migration distance is 80 mm and the whole gradient time takes 4.5 h.



▲ 25-step AMD2 gradient based on dichloromethan

## Derivatization

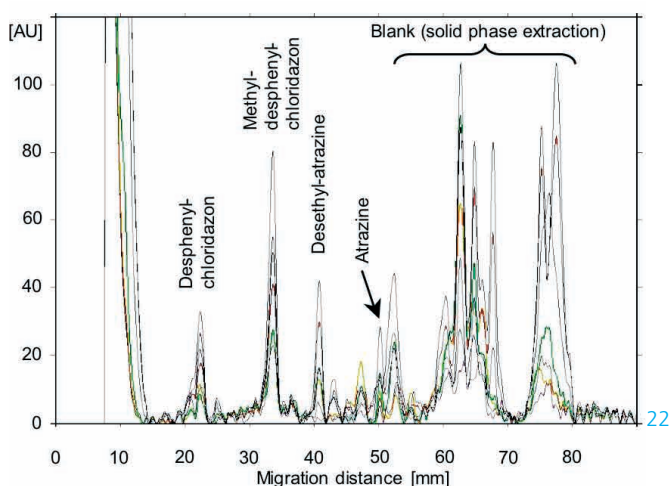
For post chromatographic derivatization the HPTLC plate is exposed to hydrochloric vapor, then to NO<sub>x</sub> for formation of the respective diazonium cations. For azo coupling the plate is dipped with the TLC immersion device III (3 cm/s, 3 s) into the Bratton-Marshall reagent (0,2 g N[1-Naphthyl]ethylenediazohydrochlorid per 100 mL methanol - dichloromethan 1:4, v/v). Colored azo dyes, i.e. red-violet zones on a pale background, result from primary aromatic amines. For stabilization of the azo dyes the plate can be exposed to ammonia.

## Densitometric evaluation

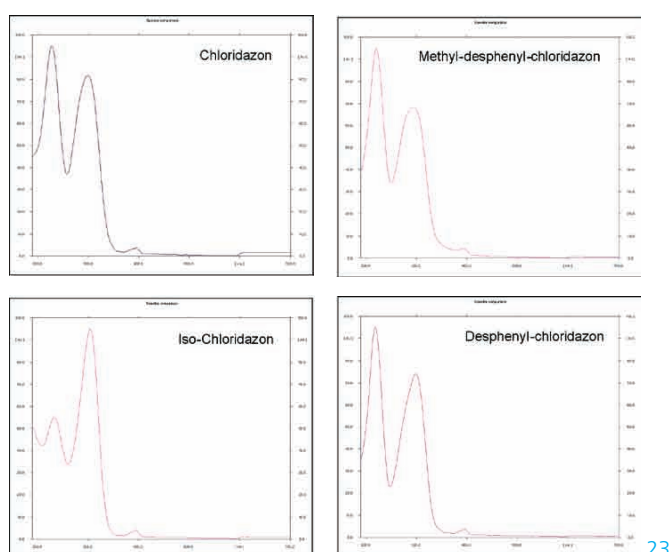
Absorbance measurement (multi-wavelength scan) in the UV range from 190 to 300 nm and in the visible range from 350 to 550 nm, respectively (after post-chromatographic derivatization) with the TLC Scanner 3 and winCATS software. Additionally UV/VIS spectra of the analytes can be recorded with the TLC Scanner 3 for spectra comparison with the standard substances (spectra identity).

## Results and discussion

During the investigation of the raw waters with HPTLC/AMD after solid phase extraction two unknown contaminants were found in addition to atrazine and desethyl-atrazine. The UV spectra of these contaminants showed exceptional local maximum wavelengths of 280/300 nm (maximum at 220 nm) and the migration distances were 22.0 mm and 33.5 mm, respectively. At first, it was not possible to correlate the spectroscopic data to known pesticides or other environmental pollutants.

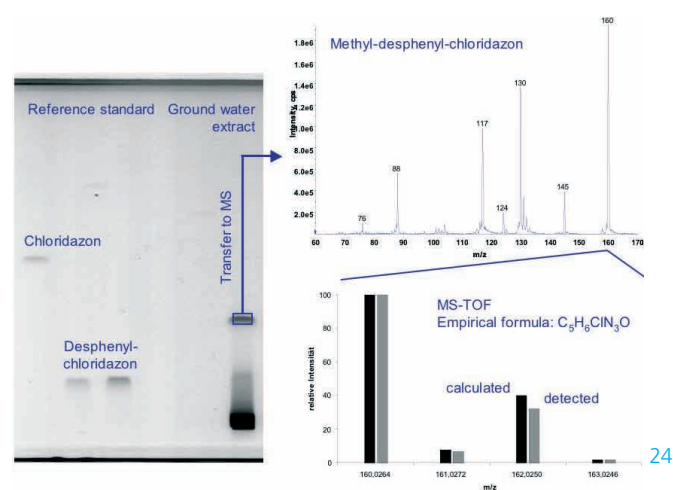


▲ HPTLC/AMD-UVVIS chromatogram of a ground water extract recorded in the multi-wavelength scan



▲ UVVIS spectra of chloridazon, iso-chloridazon, desphenyl-chloridazon and methyl-desphenyl-chloridazon (190–700 nm)

For structure elucidation the unknown substances were transferred to a time-of-flight mass spectrometer by the extraction device ChromeXtract®. The molecular weights of 145.0042 g/mol and 159.0199 g/mol were detected, whereas the difference of 14 amu indicated a methyl group and the isotopic pattern showed the presence of monochloro compounds. In conjunction with the detection of primary amino groups it was possible to identify the metabolites desphenyl-chloridazon and methyl-desphenyl-chloridazon.



▲ Mass spectra of the second unknown methyl-desphenyl-chloridazon after transfer from the HPTLC plate (before derivatization with the Bratton-Marshall reagent)

Furthermore, the removal of chloridazon and desphenyl-chloridazon by ozonation was investigated in bench-scale experiments. Here, aqueous unbuffered solutions with concentrations of 10 µg/L and 0.5 µg/L, respectively, were used. An ozone-air mixture was continuously added to a semi-batch reactor while stirring the test solution. The maximum ozone concentration was 3 mg/L. The investigations revealed that the oxidation rate of desphenyl-chloridazon was lower compared to the rate of chloridazon. After 6 min contact time chloridazon was fully oxidized, whereas desphenyl-chloridazon was not degraded until a contact time of 8 min. Likely, the attack of ozone to the phenyl group of chloridazon leads to the higher oxidation rate during ozonation.

Overall, the present investigations within the scope of a monitoring program in South Germany in-

icated that probably the herbicide metabolite desphenyl-chloridazon can be found in many other regions as well. The metabolite was found very frequently and in relatively high concentrations in surface water and ground water.



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### **CAMAG AMD 2 System** (Automated Multiple Development)

Again it was AMD, which found two unknown contaminants during the routine investigation of the water resources within the scope of the "multidimensional screening".

Besides Bayer Industry Services (see CBS 96, p. 2–5) also the laboratory for operation control and research of Zweckverband Landeswasserversorgung employs AMD due to the enhanced separation performance in combination with the multi-wavelength scan of the winCATS software to screen in a more comprehensive way.

A high degree of automation makes the planar chromatographic method very competitive. AMD is used when the desired resolution is unattainable over the available separation distance by one step isocratic development. This is often the case for complex samples with high or differing matrix content, mixtures of components with a wide polarity range, or for multi-component mixtures. The combination of multiple and gradient development leads to a focusing effect of the zones and peak sharpness is improved. This often leads to an increased sensitivity of detection.

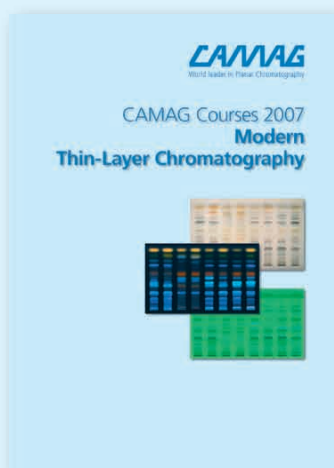
Further information is available from the authors on request.

\* Dr. W. Weber, Zweckverband Landeswasserversorgung, Betriebs- und Forschungslaboratorium, Am Spitzigen Berg 1, D-89129 Langenau, Germany, weber.w@lw-online.de

[1] W. Weber, W. Seitz, W. Schulz, H.-A. Wagener, Vom Wasser 105 (1) (2007)

# Education and training in planar chromatography

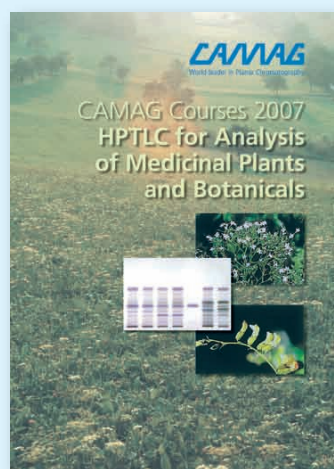
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27

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